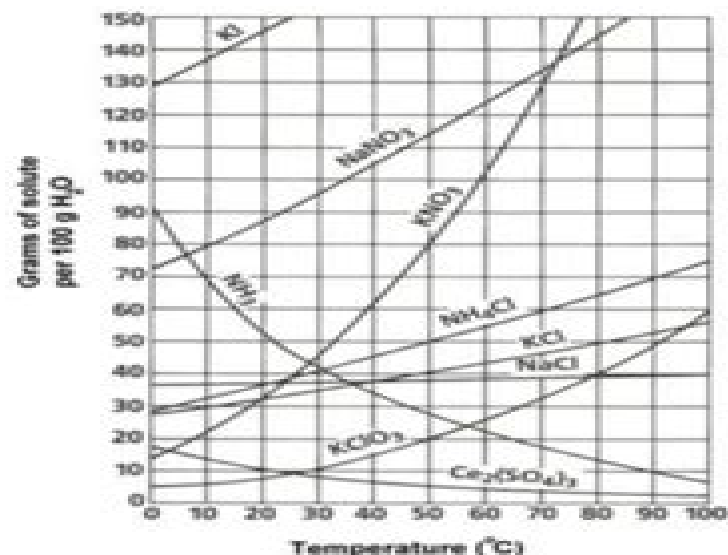


Chemistry I-H
Solubility Curves and Solutions Review Sheet
Solution Set

1. Define solubility: **The maximum amount of solute that dissolves in a fixed quantity of a particular solvent at a specified temperature.**

Use the solubility curve below to answer questions #2 - #15. Be sure to note the units on the axes of the graph.



2. In general, how does temperature affect solubility?
The higher the temp. the more substance dissolves
3. **KClO₃** Which compound is least soluble at 10°C?
4. **50 g** How many grams of KCl can be dissolved in 100g of water at 80°C?
5. **40 g** How many grams of NaCl can be dissolved in 100g of water at 90°C?
6. **about 60 g** At 40°C, how much KNO₃ can be dissolved in 100g of water?
7. **NaCl** Which compound shows the least amount of change in solubility from 0°C-100°C?
8. **unsaturated** At 30°C, 90g of NaNO₃ is dissolved in 100g of water. Is this solution saturated or unsaturated?
9. **D** At 60 C, 72 g of NH₄Cl are dissolved in 100g of water. This solution is considered to be:
A) saturated; B) unsaturated; C) supersaturated; D) saturated with some left undissolved
10. **30 g** A saturated solution of KClO₃ is formed from one hundred grams of water. If the saturated solution is cooled from 90°C to 50°C, how many grams of precipitate are formed?
11. **20 g** A saturated solution of NH₄Cl is formed from one hundred grams of water. If the saturated solution is cooled from 80°C to 40°C, how many grams of precipitate are formed?
12. **E** Which compounds show a *decrease* in solubility from 0°C-100°C?
A) potassium chlorate B) cerium sulfate C) sodium nitrate D) ammonia E) Both B and D
13. **C** Which compound is the most soluble at 10°C?
A) Ce₂(SO₄)₃ B) NaCl C) KCl D) KClO₃ E) NH₃
14. **C** A chemist adds 120 grams of KCl into a test tube which contains 100 grams of water. The temperature of the water is 80°C. How much of the solute will be left undissolved at the bottom of the test tube?
A) 50 grams B) 60 grams C) 70 grams D) 80 grams E) 56 grams

Solutions Review Worksheet Answers

**Cheung, Lillian, Dart, Hank, Kalin,
Sari, Otis, Brett, Gortmaker, Steven**

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Logic In Wonderland: An Introduction To Logic Through Reading Alice's Adventures In Wonderland - Teacher's Guidebook Nitsa Movshovitz-hadar,Atara Shriki,2018-10-08 This guidebook is for college instructors who teach a course in Introduction to Logic at a teachers college or provide a workshop in this subject for in service mathematics teachers It can also be used by high school mathematics teachers for teaching students who are capable and interested in Logic Learning is based on reading Alice s Adventures in Wonderland and discussing quotes from that book as a trigger for developing basic notions in Logic This guidebook includes the student s worksheets with exemplary solutions the background in elementary logic and pedagogical comments There is a student s workbook that accompanies this guidebook which includes the student s worksheets without solutions Ordinary textbooks for such a course are purely mathematical in their

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Intermediate Algebra Raymond A. Barnett,Thomas J. Kearns,1986 *Making Math Accessible for the At-Risk Student* Linda Lee Ptacek,2011-01-14 This invaluable collection of activities and strategies will empower teachers to help students who are struggling with math Every day secondary math teachers face classrooms containing students with a wide range of abilities yet each child is expected to meet the same testing standards Special education teachers are often asked to collaborate in classrooms outside of their curricular areas providing accommodations and modifications Both math teachers and special education instructors can benefit from effective alternative presentation strategies specifically designed for students struggling with math Making Math Accessible for the At Risk Student comprises organizational instructional and

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