

SOLUBILITY CURVES

Answer the following questions based on the solubility curve below.

Name

Key

WJCT

- Which salt is least soluble in water at 20° C? KClO₃
- How many grams of potassium chloride can be dissolved in 200 g of water at 80° C?
50g x 2 = 100g
- At 40° C, how much potassium nitrate can be dissolved in 300 g of water? 41g x 3 = 123g
- Which salt shows the least change in solubility from 0° - 100° C?

NaCl

- At 30° C, 90 g of sodium nitrate is dissolved in 100 g of water. Is this solution saturated, unsaturated or supersaturated?

unsaturated

- A saturated solution of potassium chlorate is formed from one hundred grams of water. If the saturated solution is cooled from 80° C to 50° C, how many grams of precipitate are formed? 20g 40g 20g

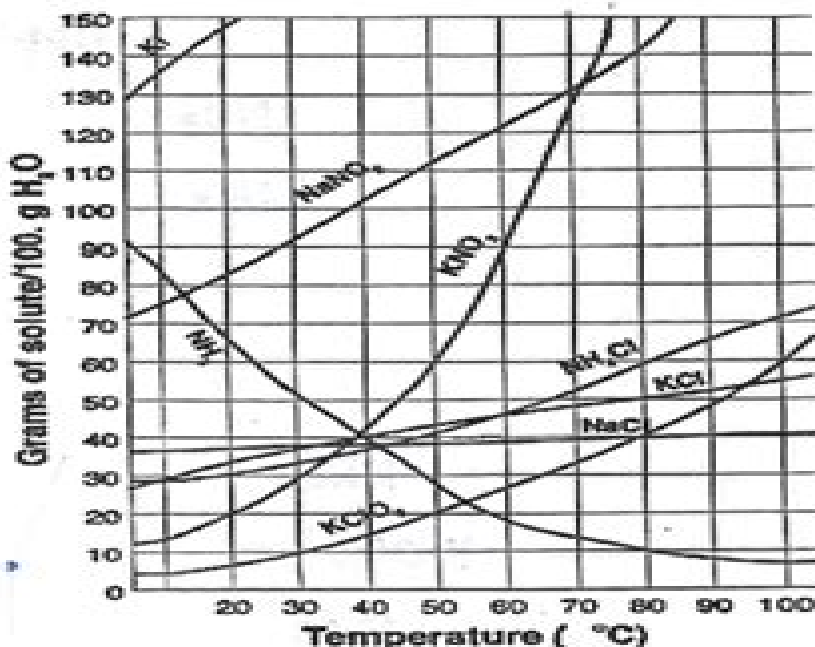
- What compound shows a decrease in solubility from 0° to 100° C? NH₃ (g)

- Which salt is most soluble at 10° C? KI

- Which salt is least soluble at 50° C? KClO₃

- Which salt is least soluble at 90° C? NH₃ → not a salt

ANSWER: NaCl



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