Write the equilibrium expression in terms of partial pressures (Ka) for each of the following reactions.

Rate the reactions in order of their increasing tendency to proceed toward completion:

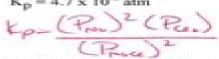
(a)  $4NH_3(g) + 3O_2(g) \Rightarrow 2N_2(g) + 6H_2O(g)$  $K_D = 1 \times 10^{228}$  atm

$$K_{p} = \frac{\left(P_{N_{2}}\right)^{2} \left(P_{N_{2}}\right)^{2}}{\left(P_{N_{2}}\right)^{3} \left(P_{0}\right)^{3}}$$
(b)  $N_{2}(g) + O_{2}(g) = 2NO(g)$ 

 $K_p = 5 \times 10^{-31}$ 

(c)  $2HF(g) \Rightarrow H_2(g) + F_2(g)$   $K_p = 1 \times 10^{-13}$ 

(d)  $2NOCl(g) \Rightarrow 2NO(g) + Cl_2(g)$  $K_p = 4.7 \times 10^{-6} \text{ atm}$ 



A Question That You Should Be Able To Answer: Why don't the Ko's in (b) and (c) have units?

7. (a) Write the K<sub>c</sub> expression for  $2 SO_{1}(g) + O_{2}(g) = 2 SO_{2}(g)$ 

Calculate the value of K.:

$$[SO_2] = 1.50 \text{ M}$$
  
 $[O_2] = 1.25 \text{ M}$   
 $[SO_3] = 3.50 \text{ M}$ 

(b) If we reverse the equation, it is:

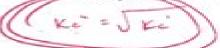
$$2 SO_3(g) = 2 SO_2(g) + O_2(g)$$

Write the K<sub>e</sub> expression for this equation and calculate the new value of Ka:

How does the expression and the value of K, in

(c) If we now multiply all of the coefficients by %:

How do they conspare with 7(b)?



(d) What would happen to the K<sub>e</sub> expression and its value if we doubled the coefficients?

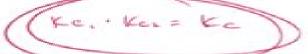
#### Sammarize:

Equation	K <sub>c</sub> expression & Value
doubled	SQUALE
reversed	in wase
halved	sa nari

Consider an equilibrium that occurs in two steps:

$$H_2S(aq) \stackrel{L}{=} H'(aq) + HS'(aq)$$
  
 $HS'(aq) \stackrel{L}{=} H'(aq) + S^2(aq)$   
 $H_2S \stackrel{L}{=} 2U+ - S^2$ 

- (a) Write the overall reaction.
- (b) How do the K<sub>c</sub>'s for the two steps (K<sub>c1</sub> & K<sub>c2</sub>) relate to the K, of the overall reaction (K<sub>c</sub>)?



# **South Pasadena Chemistry Answers**

**Camilla Rothe** 

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In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**South Pasadena Chemistry Answers**," an enthralling opus penned by a very acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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