

Questions 11 and 12 are based on the accompanying graph.

USE THIS SPACE FOR FIGURING.

1

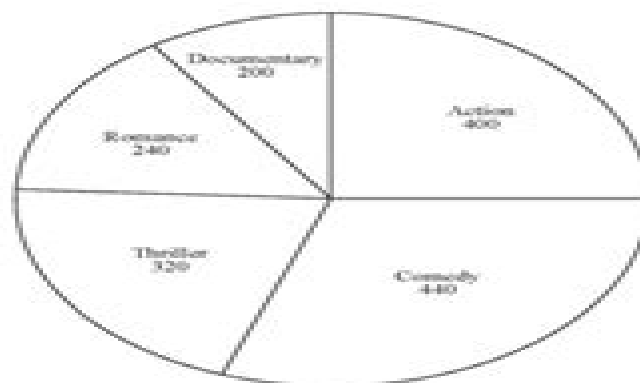
11. What fraction of the people chose Documentary?

(A) $\frac{3}{8}$
(B) $\frac{1}{4}$
(C) $\frac{1}{5}$
(D) $\frac{3}{20}$
(E) $\frac{1}{8}$

12. The number of people who chose Romance is what percent of the number of people who chose Thriller?

(A) 25%
(B) 50%
(C) 67%
(D) 75%
(E) 90%

FAVORITE MOVIE GENRE



13. A particular entrance is 84 inches tall. Kyle is planning a costume party and wants to enter the room through this doorway. If Kyle is 1.5 meters tall, the tallest hat he can place on his head and still be able to walk through the doorway without his hat touching the frame of the entrance must be less than which of the following?

(1 m = 39 in)
(A) 24 in
(B) 25 in
(C) 25.5 in
(D) 39 in
(E) 58.5 in

GO ON TO THE NEXT PAGE.

Ssat Upper Level Practice Test

Minjie Lin



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web mar 27 2023 a the lewis structure for water is represented by the given option a in this structure the oxygen atom is in the center and is bonded to two hydrogen atoms the oxygen atom has two lone pairs of electrons this structure follows the octet rule where each atom has a full valence shell with eight electrons

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web lewis structures represent the valence electrons present in the molecule as lone pairs unshared electrons drawn as dots or bonds shared electrons drawn as lines procedure for drawing lewis structures add up the total number of valence electrons that each atom contributes to the molecule polyatomic ion

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web the pcl 3 lewis structure has the typical case of phosphorus p in the center with 3 bonds to 3 other atoms phosphorus is from the same column as nitrogen in the periodic table meaning that p and n generally have the same bonding structure note the lone pair dots without bonds on top of p just like for n in the previous example for nh 3 chlorine cl

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web step 1 figure out how many electrons the molecule must have based on the number of valence electrons in each atom when drawing the structure of an ion be sure to add subtract electrons to account for the charge step 2 connect the atoms to each other with single bonds to form a skeleton structure

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web this exercise required us to take the formulas in the data table and calculate the number of valence electrons using the periodic table draw its lewis structure on paper build the molecular model and record a photo list the number of bonds and lone pairs surrounding the central atom and lastly identify the structure geometry

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