## HW#5: Solid State Physics

Micah Laing Joe Jensen PHYS-3120 2024-03-05

## 1. Electrons in Copper

What is the de Broglie wavelength of electrons in copper?

The Fermi-energy of electrons in copper is  $E_F = 7.0354$  eV and the de Broglie wavelength is given by the equation:

$$\lambda = \frac{h}{\sqrt{2m_e E_F}}$$

he|-|-- h = Quantity[4.13567 \* 10^-15, "flectronwolts" "Seconds"];

me = Quantity[0.511 / c^2, "Megaelectronwolts"];

ef = Quantity[7.0354, "flectronwolts"];

c = Quantity[3 \* 10^8, "Meters" "Seconds"^-1];

v1 = h / Sqrt[2 \* me \* ef]

4.62698 × 10^-1 =

The de Broglie wavelength of electrons in copper is 0.463 nm.

Compare the de Broglie wavelength to the atomic separation in copper.

The density of copper is 8.94 g·cm<sup>-3</sup> and its mass is 63.546 g·mol<sup>-1</sup>.

We can find the atomic separation of copper by:

$$d = \left(\frac{63,546 \, g}{\text{mol}} \times \frac{\text{cm}^3}{8.94 \, g} \times \frac{1 \, \text{mol}}{6.022 \cdot 10^{23}}\right)^{1.0}$$

# **Ssphysics 3 Term Scheme**

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