

A wide-angle photograph of the interior of St John's Co-Cathedral in Valletta, Malta. The image captures the vastness of the nave, with its high, vaulted ceiling adorned with intricate gold leaf decorations and frescoes. The floor is covered in a complex mosaic of square tiles, each featuring a different religious scene or figure. In the center, the altar is elevated on a series of steps and is surrounded by a canopy. The pews are arranged in rows, and the overall atmosphere is one of grandeur and historical significance.

A BRIEF GUIDE TO
St John's Co-Cathedral
AND ITS TREASURES · VALLETTA, MALTA

St Johns Cathedral Travel Guide

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web we can balance it for charge by adding two electrons to the right side of the equation so that the net charge on each side is 0 oxidation $\text{Ni} \rightarrow \text{Ni}^{2+} + 2\text{e}^-$ now that the oxidation half reaction is balanced it tells us that two electrons are produced for every atom of nickel oxidized but where do those electrons go

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web in the redox equation below that occurs in our stomach which of the following pairs identifies the reducing and oxidizing agents respectively $\text{C}_6\text{H}_8\text{O}_6 + 2\text{HNO}_2 \rightarrow \text{C}_6\text{H}_6\text{O}_6 + 2\text{H}_2\text{O} + 2\text{NO}$ choose 1 answer $\text{C}_6\text{H}_8\text{O}_6$ and NO_2 a $\text{C}_6\text{H}_8\text{O}_6$ and NO_2 no 2 and $\text{C}_6\text{H}_8\text{O}_6$ b NO_2 and $\text{C}_6\text{H}_8\text{O}_6$ h and no 2 c h and