

- i) Suggest a suitable substance which could be used for the electrodes.

_____ 1mark

- ii) State the name of the gas given off at the electrode B?

_____ 1mark

- b) Write the reaction that takes place at each electrode when copper sulphate solution is electrolysed using :

- i) Graphite electrodes

- ii) Copper electrodes.

(4marks)

Electrode	Solution	Reaction takes place at the anode	Reaction takes place at the cathode.
Graphite electrode	Copper sulphate solution		
Copper electrode	Copper sulphate solution		

- c) What mass of metallic silver will be deposited when a current of 0.53A is passed through a silver nitrate solution for 95 minutes? (mass Ag = 108, N = 14, O = 16)
(4marks)

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Table of Contents Sssscheme Of Work 4 Chemistry 2015 16

1. Understanding the eBook Sssscheme Of Work 4 Chemistry 2015 16
 - The Rise of Digital Reading Sssscheme Of Work 4 Chemistry 2015 16
 - Advantages of eBooks Over Traditional Books
2. Identifying Sssscheme Of Work 4 Chemistry 2015 16
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Sssscheme Of Work 4 Chemistry 2015 16
 - User-Friendly Interface
4. Exploring eBook Recommendations from Sssscheme Of Work 4 Chemistry 2015 16
 - Personalized Recommendations
 - Sssscheme Of Work 4 Chemistry 2015 16 User Reviews and Ratings
 - Sssscheme Of Work 4 Chemistry 2015 16 and Bestseller Lists
5. Accessing Sssscheme Of Work 4 Chemistry 2015 16 Free and Paid eBooks
 - Sssscheme Of Work 4 Chemistry 2015 16 Public Domain eBooks
 - Sssscheme Of Work 4 Chemistry 2015 16 eBook Subscription Services

- Sssscheme Of Work 4 Chemistry 2015 16 Budget-Friendly Options
- 6. Navigating Sssscheme Of Work 4 Chemistry 2015 16 eBook Formats
 - ePub, PDF, MOBI, and More
 - Sssscheme Of Work 4 Chemistry 2015 16 Compatibility with Devices
 - Sssscheme Of Work 4 Chemistry 2015 16 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Sssscheme Of Work 4 Chemistry 2015 16
 - Highlighting and Note-Taking Sssscheme Of Work 4 Chemistry 2015 16
 - Interactive Elements Sssscheme Of Work 4 Chemistry 2015 16
- 8. Staying Engaged with Sssscheme Of Work 4 Chemistry 2015 16
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Sssscheme Of Work 4 Chemistry 2015 16
- 9. Balancing eBooks and Physical Books Sssscheme Of Work 4 Chemistry 2015 16
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Sssscheme Of Work 4 Chemistry 2015 16
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Sssscheme Of Work 4 Chemistry 2015 16
 - Setting Reading Goals Sssscheme Of Work 4 Chemistry 2015 16
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Sssscheme Of Work 4 Chemistry 2015 16
 - Fact-Checking eBook Content of Sssscheme Of Work 4 Chemistry 2015 16
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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