

Griddable Science Items

Topography map

Density

Atoms

$$F = ma$$

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0	0	0	0		0	0
1	1	1	1		1	1
2	2	2	2		2	2
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Designed for

8th grade STAAR Test



but would work for other standardized tests

Crackables

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1. A balance and a graduated cylinder are used to determine the density of a mineral sample. The sample has a mass of 75.1 g and a volume of 14.2 cm³. What is the density of the mineral sample in g/cm³?

$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{75.1}{14.2} = 5.3 \text{ g/cm}^3$$

2. 1 mL = 1 cm³

What is the density of the metal in g/cm³?

$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{100}{20} = 5.0 \text{ g/cm}^3$$

3. Several glass cylinders about 10 cm in diameter, for use in the problems, are shown to use the

Name: **ANSWERS**

1. What is the density of the liquid in the graduated cylinder?

$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{100}{25} = 4.0 \text{ g/cm}^3$$

2. The sample of metal has a mass of 80 g. What is the density in g/cm³?

$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{80}{20} = 4.0 \text{ g/cm}^3$$

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Show What You Know Publishing

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the 'Show What You Know Publishing' text.

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