Quizizz	NAME:	
AP ChemistryStoichiometry Plus 15 Questions	CLASS:	
	DATE :	

15	Questions	D	ATE:		
1.	What is the percent by mass of oxy	gen in Mg(>?		
Α.	50%	В	60%		
C.	20%	D	40%		
2.	How many molecules are in 2.5 mol of NaCP				
Δ,	4.15 molecules	В	1.51 x10 ²⁴ molecules		
C	1.50 molecules	D	1.50 x10 ²⁴ molecules		
3.		etermined	nethane (CH ₄) present in a 100g sample of that there's 0.6 moles of methane present n of methane in the sample?		
Α,	9.6%	B	.096%		
C	.9%	D	90%		
4	A 15.67 g sample of a hydrate of magnesium carbonate was heated to drive off the water. The mass was reduced to 7.58 g. What is the formula of the hydrate?				
Α.	MgCO ₃ · 5H ₂ O	В	MgCO ₃ · 4H ₂ O		
C	MgCO ₃ · 6H ₂ O	D	MgCO ₃ · 1H ₂ O		
5.	What is the empirical formula of a c hydrogen?	ompound	that is 81.82% carbon and 18.18%		
Α,	C ₄ H ₁₀	В	C_2H_2		
C	CH ₄	D	C ₃ H ₈		

2 KClO₃ → 2 KCl + 3 O₂
 How many moles of oxygen are produced when 6.7 moles of KClO₃ decompose completely?

Star Test 10th Grade Chemistry

RJ Shavelson

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Lubricant Additives Leslie R. Rudnick,2017-07-12 This indispensable book describes lubricant additives their synthesis chemistry and mode of action All important areas of application are covered detailing which lubricants are needed for a particular application Laboratory and field performance data for each application is provided and the design of cost effective environmentally friendly technologies is fully explored This edition includes new chapters on chlorohydrocarbons foaming chemistry and physics antifoams for nonaqueous lubricants hydrogenated styrene diene viscosity modifiers alkylated aromatics and the impact of REACh and GHS on the lubricant industry

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