

4. A solution contains 2.0 g of NaCl in 100 mL of water. The solution is heated to 100°C, and the water is evaporated. How much NaCl is left?
 - a. 2.0 g
 - b. 10.0 g
 - c. 10.2 g
 - d. 10.5 g
5. A solution contains 2.0 g of NaCl in 100 mL of water. The solution is heated to 100°C, and the water is evaporated. How much NaCl is left?
 - a. 2.0 g
 - b. 10.0 g
 - c. 10.2 g
 - d. 10.5 g
6. A solution contains 2.0 g of NaCl in 100 mL of water. The solution is heated to 100°C, and the water is evaporated. How much NaCl is left?
 - a. 2.0 g
 - b. 10.0 g
 - c. 10.2 g
 - d. 10.5 g
7. A solution contains 2.0 g of NaCl in 100 mL of water. The solution is heated to 100°C, and the water is evaporated. How much NaCl is left?
 - a. 2.0 g
 - b. 10.0 g
 - c. 10.2 g
 - d. 10.5 g
8. A solution contains 2.0 g of NaCl in 100 mL of water. The solution is heated to 100°C, and the water is evaporated. How much NaCl is left?
 - a. 2.0 g
 - b. 10.0 g
 - c. 10.2 g
 - d. 10.5 g
9. A solution contains 2.0 g of NaCl in 100 mL of water. The solution is heated to 100°C, and the water is evaporated. How much NaCl is left?
 - a. 2.0 g
 - b. 10.0 g
 - c. 10.2 g
 - d. 10.5 g
10. A solution contains 2.0 g of NaCl in 100 mL of water. The solution is heated to 100°C, and the water is evaporated. How much NaCl is left?
 - a. 2.0 g
 - b. 10.0 g
 - c. 10.2 g
 - d. 10.5 g

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