

Name _____
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2 • Atomic Structure & Nuclear Chemistry

ATOMIC NUMBER & MASS NUMBER

Complete the following chart and answer the questions below. *HINT: The number that appears after the element name in the first column is the mass number.*

	<i>Element Name</i>	<i>Atomic Number</i>	<i>Number of Protons</i>	<i>Number of Neutrons</i>	<i>Mass Number</i>
1.	carbon – 12				12
2.		8		8	
3.	hydrogen – 1				1
4.			6		14
5.	hydrogen – 3			2	
6.	nitrogen – 14				14
7.				1	2
8.		92		146	
9.	cesium – 137			82	
10.		11		12	
11.			47		108
12.	tungsten – 184			110	
13.				45	80
14.			24		52
15.				89	152
16.	silver – 107				107
17.		76		114	

18. How are the *atomic number* and the *number of protons* related to each other?

19. How do the *number of protons*, *number of neutrons*, and the *mass number* relate to each other?

20. What is the *one thing* that determines the identity of an atom (that is, whether it is an oxygen atom or a carbon atom, etc.)?

Skill Practice 16 Atomic Size Practice Answers

Terry C. Jones



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Table of Contents Skill Practice 16 Atomic Size Practice Answers

1. Understanding the eBook Skill Practice 16 Atomic Size Practice Answers
 - The Rise of Digital Reading Skill Practice 16 Atomic Size Practice Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Skill Practice 16 Atomic Size Practice Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Skill Practice 16 Atomic Size Practice Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Skill Practice 16 Atomic Size Practice Answers
 - Personalized Recommendations
 - Skill Practice 16 Atomic Size Practice Answers User Reviews and Ratings
 - Skill Practice 16 Atomic Size Practice Answers and Bestseller Lists
5. Accessing Skill Practice 16 Atomic Size Practice Answers Free and Paid eBooks
 - Skill Practice 16 Atomic Size Practice Answers Public Domain eBooks
 - Skill Practice 16 Atomic Size Practice Answers eBook Subscription Services
 - Skill Practice 16 Atomic Size Practice Answers Budget-Friendly Options

6. Navigating Skill Practice 16 Atomic Size Practice Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Skill Practice 16 Atomic Size Practice Answers Compatibility with Devices
 - Skill Practice 16 Atomic Size Practice Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Skill Practice 16 Atomic Size Practice Answers
 - Highlighting and Note-Taking Skill Practice 16 Atomic Size Practice Answers
 - Interactive Elements Skill Practice 16 Atomic Size Practice Answers
8. Staying Engaged with Skill Practice 16 Atomic Size Practice Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Skill Practice 16 Atomic Size Practice Answers
9. Balancing eBooks and Physical Books Skill Practice 16 Atomic Size Practice Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Skill Practice 16 Atomic Size Practice Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Skill Practice 16 Atomic Size Practice Answers
 - Setting Reading Goals Skill Practice 16 Atomic Size Practice Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Skill Practice 16 Atomic Size Practice Answers
 - Fact-Checking eBook Content of Skill Practice 16 Atomic Size Practice Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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