Dutte

Skills Worksheet

Active Reading

Section: Changing Population Trends

Read the passage below and answer the questions that follow.

In many of the poorest countries, wood is the main fuel source. When populations are stable, people use fallen tree limbs for fuel, which does not harm the trees. When populations grow rapidly, deadwood does not accumulate fast enough to provide enough fuel. People begin to cut down living trees, which reduces the amount of wood available in each new year. Parts of Africa, Asia, and India have been cleared of vegetation by people collecting fuel wood.

A supply of fuel ensures that a person can boil water and cook food. In many parts of the world, water taken directly from wells or public supplies is not safe to drink because it may carry waterborne parasites or other diseases. The water can be sterilized by boiling it, but fuel is needed to do so. Also, food is often unsafe or harder to digest unless it is cooked. Without enough fuelwood, many people suffer from disease and malnutrition.

IDENTIFYING MAIN IDEAS

One reading skill is the ability to identify the main ideas of a passage. The main idea is the main focus or key idea. Frequently, a main idea is accompanied by supporting information that offers detailed facts about main ideas.

Read each question and write the answer in the space provided.

- 1. The main source of fuel for many poorer countries is
- 2. People use fallen tree limbs for fuel when their population is
- 3. If a population grows quickly, people begin to cut down for fuel wood.
- Public water supplies are unsafe in some parts of the world because the water may carry_____ and _____.
- Explain how an area of land can become cleared of vegetation because a population grows.

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Patrick Vollmar

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web answers lewis dot lewis dot structures for each molecule on the worksheet the lewis dot structure the number of valence electrons the electron arrangement e a and the molecular geometry m g are given respectively to see a larger view of the lewis dot structure point at the molecule right click on the mouse and go to view image

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web you should try to answer the questions without referring to your textbook if you get stuck try asking another group for help for each of the following draw the lewis dot structure give the electron arrangement e a and the molecular geometry m g

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web in 1916 american chemist gilbert n lewis introduced bond lines to electron dot structures these structures also known as lewis structures or electron dot structures are drawings that visually demonstrate how electrons are shared and arranged around atoms the electrons denoted as dots are called lone pairs and belong to an individual

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web lewis dot structures are useful for describing chemical bonds but have some flaws a lewis structure is a picture of a molecule that shows the covalent bonds and pairs of free electrons the octet rule is the basis for lewis structures

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web lewis dot formula also called an electron dot formula shows the valence electrons indicating the bonding between atoms the following guidelines will help draw the electron dot formulas correctly

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web oct 29 2021 contributors and attributions learning objectives to draw lewis structures for molecules and polyatomic ions with one central atom introduction to lewis structures a lewis structure is a way to show how atoms share electrons when they form a molecule lewis structures show all of the valence electrons in an atom or molecule

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web lewis structures we also use lewis symbols to indicate the formation of covalent bonds which are shown in lewis structures drawings that describe the bonding in molecules and polyatomic ions for example when two chlorine atoms form a chlorine molecule they share one pair of electrons

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web this online quiz is intended to give you extra practice in identifying and drawing lewis dot structures as well as predicting ion formation this quiz aligns with the following ngss standard s hs ps1 1

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web jan 30 2023 a lewis structure is a very simplified representation of the valence shell electrons in a molecule it is used to show how the electrons are arranged around individual atoms in a molecule electrons are shown as dots or for bonding electrons as a line between the two atoms

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web jan 30 2023 follow these simple steps to draw lewis dot structures draw the atoms on paper and put dots around them to represent valence electrons of the atom be sure to have the correct number of electrons if the species is an ion add or subtract electrons corresponding to the charge of the ion

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Skills Worksheet Active Reading 8 Answer Key

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