

BALANCING CHEMICAL EQUATIONS

Load the simulation *Balancing Chemical Equations* <http://phet.colorado.edu/en/simulation/balancing-chemical-equations>

CRITICAL THINKING QUESTIONS

- Explore the *Balancing Chemical Equations* simulation. Discuss with your group what you find.
 - What are the different ways that the simulation indicates when an equation is balanced?

The molecules will show the same # of atoms on each side.

- For each balanced reaction, indicate the total number of molecules in the table below.

Reaction	Reactant Side (Left)	Product Side (Right)
Make Ammonia	4	2
Separate Water	2	3
Combust Methane	3	3
Total Number of Molecules	9	8

- Is the number of total molecules on the left side of a balanced equation always equal to the number of total molecules on the right side of the equation? Explain your answer.

No, only the combust methane has the same number of molecules on each side.

- For each balanced reaction, indicate the total number of atoms in the table below.

Reaction	Reactant Side (Left)	Product Side (Right)
Make Ammonia	8	8
Separate Water	6	6
Combust Methane	9	9
Total Number of Atoms	23	23

- Is the number of total atoms on the left side of a balanced equation always equal to the number of total atoms on the right side of the equation?

Yes

Skill Builder Balancing Chemical Equations Answers

James Lee Merrifield



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