

Practice Problem #7



Solid silver chromate (Ag_2CrO_4) forms when potassium chromate (K_2CrO_4) is added to a solution containing .500g of silver nitrate (AgNO_3). Determine the theoretical yield of Ag_2CrO_4 . Calculate the percent yield if the reaction yields .455g of Ag_2CrO_4 .

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S Ashworth



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