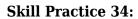
# Practice Problem #7

Solid silver chromate (AgCrO<sub>4</sub>) forms when potassium chromate (K<sub>2</sub>CrO<sub>4</sub>) is added to a solution containing .500g of silver nitrate (AgNO<sub>3</sub>). Determine the theoretical yield of Ag<sub>2</sub>CrO<sub>4</sub>. Calculate the percent yield if the reaction yields .455g of Ag<sub>2</sub>CrO<sub>4</sub>.

## **Skill Practice 34**

### **S** Ashworth



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