Name: \_\_\_\_\_

Date: \_\_\_\_\_

## **Balancing Chemical Equations**

Balance the following chemical equations.

1. 3 KOH + 1 
$$H_3PO_4$$
 -> 1  $K_3PO_4$  + 3  $H_2O$ 

2. 
$$4$$
 NH<sub>3</sub> +  $5$  O<sub>2</sub>  $\longrightarrow$   $4$  NO +  $6$  H<sub>2</sub>O

4. 
$$\underline{2}$$
 CaS +  $\underline{2}$  H<sub>2</sub>O  $\longrightarrow$   $\underline{1}$  Ca(HS)<sub>2</sub> +  $\underline{1}$  Ca(OH)<sub>2</sub>

5. 
$$\underline{1}$$
 Au<sub>2</sub>S<sub>3</sub> +  $\underline{3}$  H<sub>2</sub>  $\longrightarrow$   $\underline{2}$  Au +  $\underline{3}$  H<sub>2</sub>S

7. 
$$\underline{1}$$
 Ba<sub>3</sub>N<sub>2</sub> +  $\underline{6}$  HF  $\longrightarrow$   $\underline{3}$  BaF<sub>2</sub> +  $\underline{2}$  NH<sub>3</sub>

# **Skill Practice 29 Balancing Answers**

**Bonnie S. Mohnsen** 

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