

Abstract

Reinforcement

Placing All Your Elements on the Table

Complete this worksheet after you have finished reading the section "Grouping the Elements."

You can tell a lot about the properties of an element just by looking at the element's location on the periodic table. This worksheet will help you better understand the connections between the periodic table and the properties of the elements. Follow the directions below, and use crayons or colored pencils to color the periodic table at the bottom of the page.

1. Color the square for hydrogen yellow.
2. Color the groups with very reactive metals red.
3. Color and label the noble gases orange.
4. Color the transition metals green.
5. Using black, mark the zigzag line that shows the position of the metalloids.
6. Color the metalloids purple.
7. Use blue to color all of the nonmetals that are not noble gases.
8. Color the metals in Groups 13–16 brown.
9. Circle and label the actinides in yellow.
10. Circle and label the lanthanides in red.
11. Circle and label the alkali metals in blue.
12. Circle and label the alkaline-earth metals in purple.
13. Circle and label the halogens in green.

A blank periodic table of elements. The table is organized into rows and columns, with atomic numbers 1 through 118 labeled above the corresponding cells. The layout includes the standard periodic table structure, including the lanthanide and actinide series. Below the main table, there are two additional empty rows for writing or drawing.

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