

Gizmos are in Classlink

Student Exploration: Solubility and Temperature

Vocabulary: concentration, dissolve, homogeneous mixture, solubility, solubility curve, solute, solution, solvent

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. What happens when you stir a spoonful of sugar into hot water?
If you stir a spoonful of sugar into hot water, the sugar will dissolve in the hot water. _____
2. When sugar or another substance is **dissolved** in water, it disappears from view and forms a **homogeneous mixture** with the water, also called a **solution**.
If you can't see the sugar, how can you tell that it is there? ____ If you taste it, it tastes sweet.

3. Does sugar dissolve more easily in hot water or cold water? __ It dissolves more easily in hot water. _____

Gizmo Warm-up

A solution generally consists of two parts, a **solute** that is dissolved and a **solvent** that the solute is dissolved into. For example, sugar is a solute that is dissolved into the solvent water. In the *Solubility and Temperature* Gizmo, you will study how temperature affects how much solute will dissolve in a solution.

To begin, check that **Potassium nitrate** is selected and the **Temp.** of the water is 20 °C. Click **OK**.



1. In this solution, what is the solute? _____ Potassium Nitrate _____ What is the solvent? ____ Water _____
2. Click **Add 10 g** to mix 10 g of potassium nitrate into the water.
A. Did all of the potassium nitrate dissolve? ____ Yes _____

Teacher Guide Solubility And Temperature Gizmo Answers

M Tight



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Chemical Tests National Science Resources Center,1994-08-01 By conducting a variety of physical and chemical tests students explore some concepts basic to general chemistry physical and chemical properties and how to describe them and changes that may occur when different solids and liquids are mixed together or separated As a result students are introduced to solubility filtration evaporation crystallization and acids bases and neutrals Page 3 **Pruebas con sustancias**

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Understanding Chemistry George C. Pimentel,Richard D. Spratley,1972

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