

# TEACHING TRANSPARENCY 18

## The Periodic Table

Use with Chapter 6.

**PERIODIC TABLE OF THE ELEMENTS**

Solid: Gas: Liquid: Metal: Nonmetal: Metalloid:

Atomic number: Symbol: Name:

1	2																	18
Hydrogen 1 H 1.01	Helium 2 He 4.00																	Neon 10 Ne 20.18
Lithium 3 Li 6.94	Beryllium 4 Be 9.01																	Argon 18 Ar 39.95
Sodium 11 Na 22.99	Magnesium 12 Mg 24.31																	Krypton 36 Kr 83.80
Calcium 20 Ca 40.08	Scandium 21 Sc 44.96																	Xenon 54 Xe 131.30
Strontium 38 Sr 87.62	Titanium 22 Ti 47.88																	Radon 86 Rn 222
Barium 56 Ba 137.3	Vanadium 23 V 50.94																	
Radium 88 Ra 226	Chromium 24 Cr 52.00																	
	Manganese 25 Mn 54.94																	
	Iron 26 Fe 55.85																	
	Cobalt 27 Co 58.93																	
	Nickel 28 Ni 58.69																	
	Copper 29 Cu 63.55																	
	Zinc 30 Zn 65.39																	
	Gallium 31 Ga 69.72																	
	Germanium 32 Ge 72.64																	
	Arsenic 33 As 74.92																	
	Selenium 34 Se 78.96																	
	Bromine 35 Br 79.90																	
	Krypton 36 Kr 83.80																	
	Rubidium 37 Rb 85.47																	
	Strontium 38 Sr 87.62																	
	Yttrium 39 Y 88.91																	
	Zirconium 40 Zr 91.22																	
	Niobium 41 Nb 92.91																	
	Molybdenum 42 Mo 95.94																	
	Technetium 43 Tc 98																	
	Ruthenium 44 Ru 101.1																	
	Rhodium 45 Rh 102.9																	
	Palladium 46 Pd 106.4																	
	Silver 47 Ag 107.9																	
	Cadmium 48 Cd 112.4																	
	Indium 49 In 114.8																	
	Tin 50 Sn 118.7																	
	Antimony 51 Sb 121.8																	
	Tellurium 52 Te 127.6																	
	Iodine 53 I 126.9																	
	Xenon 54 Xe 131.3																	
	Cesium 55 Cs 132.9																	
	Barium 56 Ba 137.3																	
	Lanthanum 57 La 138.9																	
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	Gadolinium 64 Gd 157.3																	
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	Dysprosium 66 Dy 162.5																	
	Homium 67 Ho 164.9																	
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	Ytterbium 70 Yb 173.0																	
	Lutetium 71 Lu 174.9																	
	Hafnium 72 Hf 178.5																	
	Tantalum 73 Ta 180.9																	
	Tungsten 74 W 183.8																	
	Rhenium 75 Re 186.2																	
	Osmium 76 Os 190.2																	
	Iridium 77 Ir 192.2																	
	Platinum 78 Pt 195.1																	
	Gold 79 Au 197.0																	
	Mercury 80 Hg 200.6																	
	Thallium 81 Tl 204.4																	
	Lead 82 Pb 207.2																	
	Bismuth 83 Bi 208.9																	
	Polonium 84 Po 209																	
	Astatine 85 At 210																	
	Radon 86 Rn 222																	
	Francium 87 Fr 223																	
	Radium 88 Ra 226																	
	Actinium 89 Ac 227																	
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# Teaching Transparency 18 The Periodic Table

**Robert C. Smoot, Smoot, Richard G.  
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applications of Neutrosophic approaches This special issue also addresses how scientific production in Neutrosophy focuses on social issues specific to Latin American philosophy In the regional context of Latin America it is possible to state that Neutrosophic tools and knowledge are used for the identification analysis and resolution of social problems offering unique approaches or distinctive contributions to the field of Neutrosophy influenced by its cultural and philosophical context Neutrosophic science in Latin America shows a clear pattern of how scientific production addresses social problems standing out for its innovative approaches that reflect the cultural and philosophical particularities of the region This approach has allowed Neutrosophy not only to advance in theoretical terms but also to provide practical and contextually relevant solutions to social challenges This special issue compiles works presented at the conference reflecting the richness and diversity of current research in this field We hope that these articles not only contribute to the advancement of knowledge in Neutrosophic theory but also inspire new research and applications in multiple disciplines     Addison-Wesley Introduction to Physical Science Michael B. Leyden,1988     **Prentice Hall Science Explorer** Michael J. Padilla,2002     The Science Teacher ,1979 Some issues are accompanied by a CD ROM on a selected topic

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