

8.2**THE NATURE OF COVALENT BONDING****Section Review****Objectives**

- State a rule that usually tells how many electrons are shared to form a covalent bond
- Describe how electron dot formulas are used
- Predict when two atoms are likely to be joined by a double or a triple covalent bond
- Distinguish between a single covalent bond and other covalent bonds
- Describe how the strength of a covalent bond is related to its bond dissociation energy
- Describe how resonance structures explain bonding
- Identify some exceptions to the octet rule

Vocabulary

- single covalent bond
- structural formulas
- unshared pairs
- double covalent bonds
- triple covalent bonds
- coordinate covalent bond
- polyatomic ion
- bond dissociation energy
- resonance structures

Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number.

When atoms share electrons to gain the 1 configuration of a noble gas, the bonds formed are 2. A 3 pair of valence electrons constitutes a 4 covalent bond. Pairs of valence electrons that are not shared between atoms are called 5. Sometimes two or three pairs of electrons may be shared to give 6 covalent bonds. In some cases, only one of the atoms in a bond provides the pair of bonding electrons; this is a 7. 8 is required to break covalent bonds between atoms. The total energy required to break the bond between two covalently bonded atoms is known as the 9.

1. _____
2. _____
3. _____
4. _____
5. _____
6. _____
7. _____
8. _____
9. _____
10. _____

When it is possible to write two or more valid electron dot formulas for a molecule or ion, each formula is referred to as a 10.

The Nature Of Covalent Bonding Worksheet

DP Hallahan



The Nature Of Covalent Bonding Worksheet:

Adopting the Melody of Appearance: An Mental Symphony within **The Nature Of Covalent Bonding Worksheet**

In some sort of consumed by screens and the ceaseless chatter of quick connection, the melodic splendor and mental symphony created by the published term usually disappear into the back ground, eclipsed by the constant noise and distractions that permeate our lives. But, set within the pages of **The Nature Of Covalent Bonding Worksheet** a stunning literary value filled with natural feelings, lies an immersive symphony waiting to be embraced. Crafted by a masterful composer of language, this fascinating masterpiece conducts visitors on an emotional trip, well unraveling the concealed tunes and profound affect resonating within each cautiously constructed phrase. Within the depths of the poignant evaluation, we can explore the book is key harmonies, analyze their enthralling publishing model, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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