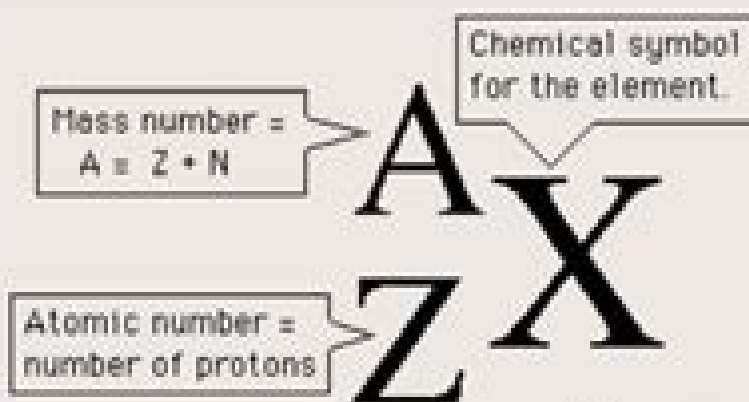


Isotope Notation



N = neutron number

	<u>Protons</u>	<u>Neutrons</u>	<u>Electrons</u>
$^{12}_6\text{C}$	6	6	6
$^{13}_6\text{C}$	6	7	6
$^{14}_6\text{C}$	6	8	6

How many protons, neutrons and electrons in each of the following:

	<u>protons</u>	<u>neutrons</u>	<u>electrons</u>
^{23}Na	11	12	11
^{14}N	7	7	7
^{38}Ar	18	20	18
^{35}Cl	17	18	17
$^{36}\text{Cl}^{-1}$	17	19	18
^{56}Fe	26	30	26

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