

SALT ANALYSIS

TEST FOR ANIONS

EXPERIMENT	OBSERVATION	INFERENCE
1. Test for CO_3^{2-} → A little of the salt is treated with dil. HCl.	Brisk effervescence with the evolution of colourless odourless gas	May be CO_3^{2-} $\text{CaCO}_3 + 2\text{HCl} \rightarrow \text{CaCl}_2 + \text{CO}_2 + \text{H}_2\text{O}$
Confirmatory test :- 2. Pass the above gas through lime water.	Lime water turns milky	Presence of CO_3^{2-} is confirmed $\text{Ca(OH)}_2 + \text{CO}_2 \rightarrow \text{CaCO}_3 + \text{H}_2\text{O}$
Test for Halides 3. A little of the salt is treated with 1 or 2 drops of conc. H_2SO_4	a) Colourless pungent smelling gas evolves, dense white fumes when a glass rod is dipped in NH_4OH is shown on the mouth of the test tube b) Reddish brown vapours with pungent smell c) Deep violet vapours with pungent smell	a) May be Cl^- $\text{NaCl} + \text{H}_2\text{SO}_4 \rightarrow \text{NaHSO}_4 + \text{HCl}$ b) Maybe Br^- $\text{NaBr} + \text{H}_2\text{SO}_4 \rightarrow \text{NaHSO}_4 + \text{HBr}$ $\text{H}_2\text{SO}_4 + 2\text{HBr} \rightarrow \text{SO}_2 + \text{Br}_2 + 2\text{H}_2\text{O}$ c) May be I^- $\text{KI} + \text{H}_2\text{SO}_4 \rightarrow \text{KHSO}_4 + \text{HI}$ $\text{H}_2\text{SO}_4 + 2\text{HI} \rightarrow \text{SO}_2 + \text{I}_2 + 2\text{H}_2\text{O}$
Confirmatory test of halides 4. A little of the salt solution is acidified with dil. HNO_3 , boil, cool and add AgNO_3 sol.	a) Curdy white ppt. soluble in NH_4OH . b) Pale yellow ppt. sparingly soluble in NH_4OH c) Yellow ppt. insoluble in NH_4OH	a) presence of Cl^- is confirmed $\text{NaCl} + \text{AgNO}_3 \rightarrow \text{NaNO}_3 + \text{AgCl (white ppt.)}$ b) presence of Br^- is confirmed $\text{KBr} + \text{AgNO}_3 \rightarrow \text{KNO}_3 + \text{AgBr (pale yellow ppt.)}$ c) presence of I^- is confirmed $\text{KI} + \text{AgNO}_3 \rightarrow \text{KNO}_3 + \text{AgI (yellow ppt.)}$

Unknown Salt Identification

David Kirk

A red circular graphic with a gradient, appearing as a partial circle or a thick arrow pointing to the right, located to the right of the name bar.

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College Practical Chemistry V K Ahluwalia, Sunita Dhingra, Sunita Dhingra, 2005, *Salts and Their Reactions*
Leonard Dobbin, Hugh Marshall, 1904 **Advanced Inorganic Analysis - I** Mr. Ajeet Singh, 2018-05-22 This book provides in depth coverage of modern methods in inorganic analysis including advanced qualitative and quantitative techniques instrumentation and interpretation of results catering to postgraduate and research level chemistry students *Abiotic stress: molecular genetics and genomics, Volume II* Mukesh Jain, Rohini Garg, Rajeev K. Varshney, Prasanta Kumar Subudhi, EduGorilla's CBSE Class 11th Chemistry Lab Manual | 2024 Edition | A Well Illustrated, Complete Lab Activity book with Separate FAQs for Viva Voce Examination, Need an informative and well illustrated Lab Manual CBSE Class 11th Chemistry Lab Manual is here for you The Lab Manual provides comprehensive steps for guiding students through each experiment Rigorously researched content prepared by a team of educators writers editors and proofreaders CBSE Class XI Chemistry Lab Manual has properly labeled high resolution diagrams and graphs A separate section on Viva Questions has been included to aid students in their Viva examination The Lab Manual explains the complex topics through detailed illustrations and lucid language making them simple to grasp Worksheets have been provided in CBSE Class 11th Chemistry Lab Manual for doing rough work **Laboratory Manual for Chemistry Fundamentals** Phyllis Buell, James Girard, 2002-08 Physical Sciences *Edexcel A Level Chemistry Student Book 1* Andrew Hunt, Graham Curtis, Graham Hill, 2015-07-17 Exam Board Edexcel Level AS A level Subject Chemistry First Teaching September 2015 First Exam June 2016 Endorsed by Edexcel Develop and assess your students knowledge and mathematical skills throughout A Level with worked examples practical assessment guidance and differentiated end of topic questions with this Edexcel Year 1 student book Identifies the level of your students understanding with diagnostic questions and a summary of prior knowledge at the start of the Year 1 Student Book Provides support for all 16 required practicals with various activities and questions along with a Practical chapter covering procedural understanding and key ideas related to measurement Mathematical skills are integrated throughout with plenty of worked examples including notes on methods to help explain the strategies for solving each type of problem Offers plenty of practice with Test Yourself Questions to help students assess their understanding and measure progress Encourages further reading and study with short passages of extension material Develops understanding with free online access to Test yourself Answers and an Extended Glossary Edexcel A level Chemistry Year 1 Student Book includes AS level Experiments in General Chemistry Kenneth W. Whitten, Kenneth D. Gailey, 1981 General, Organic, and Biochemistry Lab Manual Sara Selfe, 2006-01-12 Offers a choice of classic chemistry experiments and innovative ones All of them place special emphasis on the biological implications of chemical concepts Available for custom publishing at <http://custompub.whfreeman.com> Lab Manual for Investigating Chemistry David Collins (Ph. D.), Matthew Johll, 2008-12-02 While many of the core labs from the first edition have been retained a renewed focus on the basics of chemistry and the scientific process

create an even more detailed supplemental offering **Modern Experimental Chemistry** George W. Jr. Latimer, 2012-12-02 Modern Experimental Chemistry provides techniques of qualitative analysis that reinforce experiments on ionic equilibria. This book includes the determination of water in hydrated salts, identification of an organic compound after determining its molecular weight and nonaqueous titration of a salt of a weak acid. The calculation of chemical stoichiometry, calculation of thermodynamic properties by determining the change in equilibrium with temperature and chromium chemistry are also covered. This compilation contains enough experiments for classes which have six hours of laboratory two 3 hour meetings per week to last two semesters. This publication is intended for chemistry students as an introductory manual to chemistry laboratory Lab Experiments in Introductory Chemistry Phil Reedy, Donald J. Wink, Sharon Fetzer-Gislason, 2003-03-21 The manual contains laboratory experiments written specifically for the prep chem lab as well as for the general chemistry course. Available as a complete manual or custom published at <http://custompub.whfreeman.com> **Standards-Based Investigations: Forensic Science**, 2008-06-02 Introduce crime scene investigation techniques familiar from popular TV programs. The high interest science activities in this resource will grab learners' interest while improving content area literacy and critical thinking skills. Interlocking reading passages and lab activities will stimulate creativity with ideas for research projects and other presentations. Includes a Teacher Resource CD with reproducible fact sheets and lab activities. This resource is aligned to the interdisciplinary themes from the Partnership for 21st Century Skills and supports core concepts of STEM instruction. 176pp *Standards-Based Investigations Forensic Science* Jeanette Jolley, John Powrie, 2008-03-14 Provides information on various aspects of forensic science appropriate for sixth through eighth grade students and includes activities and comprehension questions that reinforce each concept. Includes CD ROM containing reproducible teacher resource materials *Sif: Chemistry 5na Wb*, **Laboratory Manual of Qualitative Analysis** Wilhelm Segerblom, 1908 **Educart 40 ICSE Class 10 Sample Papers Volume 1 - Physics Chemistry Biology and Mathematics for 2024-25 Exam (with exclusive ICSE Mock Booklets 2025)** Educart, Sir Tarun Rupani, 2024-10-19 What You Get Increased 25% Competency based Qs ICSE Mock Test Booklets Educart 40 ICSE Class 10 Sample Papers Volume 1 Physics Chemistry Biology and Mathematics for 2024-25 Exam with exclusive ICSE Mock Booklets 2025 Based on ICSE Specimen Papers released on May 15th 2024 Includes 5 previous year papers for every included subject 3 new paper sets include competency based high order ability questions 2 ICSE mock test booklets with answer writing OMR booklets Why choose this book Practice in an exam like environment with ICSE mock test booklets and score high on the boards *HORIZON OF FIELD CROPS* Ratikanta Maiti, 2017-08-14 The colossal importance of various field crops to satisfy hunger and other requirements of human beings is well known. The roles of cereals and pulses on human nutrition deserves special emphasis for billions of human populations in the world. The present books brief accounts of various aspects of important cereal crops sugarcane various legumes oil seed crops and fiber yielding crops of the world in

different chapters with illustrations It deals origin and domestication systematic positions utilization botanical description vegetative and reproductive growth physiology mineral nutrition productivity and abiotic stress resistance of most of the crops and also discusses the mechanism of tolerance to drought and salinity The book also deals with various aspects of fiber crops In the last two chapters are discussed researches undertaken on salinity tolerance of few crops Therefore the book deals in brief the major aspects of most of the field crops in the world Not a single book is available in the market dealing with so many aspects all together The book can serve as a text book in economic botany agriculture and serve the needs of researcher s working on various crops with research advances obtained on these crops **The Science of James**

Smithson Steven Turner, 2020-11-03 A quirky oddly touching book that allows us to step for a few moments inside the world of a practicing Enlightenment scientist to sit beside him as he fans the flames of a candle with his little blowpipe waiting for that small mineral in front of him to melt and yield its secrets Wall Street Journal An accessible exploration of the noteworthy scientific career of James Smithson who left his fortune to establish the Smithsonian Institution James Smithson is best known as the founder of the Smithsonian Institution but few people know his full and fascinating story He was a widely respected chemist and mineralogist and a member of the Royal Society but in 1865 his letters collection of 10 000 minerals and more than 200 unpublished papers were lost to a fire in the Smithsonian Castle His scientific legacy was further written off as insignificant in an 1879 essay published through the Smithsonian fifty years after his death a claim that author Steven Turner demonstrates is far from the truth By providing scientific and intellectual context to his work The Science of James Smithson is a comprehensive tribute to Smithson s contributions to his fields including chemistry mineralogy and more This detailed narrative illuminates Smithson and his quest for knowledge at a time when chemists still debated thing as basic as the nature of fire and struggled to maintain their networks amid the ever changing conditions of the French Revolution and the Napoleonic Wars

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