



Name: \_\_\_\_\_ Date: \_\_\_\_\_  
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**Rule 1** A number is not needed to locate the double bond in unsaturated hydrocarbons with only one double bond because that bond is assumed to be between carbons 1 and 2.



cyclohexene



cyclopentene



1,3-cyclohexadiene



1,3-cycloheptadiene

**Rule 2** In unsaturated hydrocarbons with only one double bond, the double bond carbon atoms are numbered 1 and 2 in the direction (priority or counterclockwise) that gives the first substituent the lower number. Again, no number is used in the name to locate the double bond.



cyclohexene



1-methylcyclopentene



1-bromocyclohexene



1-ethyl-1-methylcyclohexene

**Rule 3** In hydrocarbons with more than one double bond within the ring, assign one double bond the numbers 1 and 2 and the other double bond the lower-numbered position.



1,3-cyclohexadiene



1,3-cycloheptadiene

### Alkenes as Substituents

An alkene group in a compound is represented by a substituent symbol in which a carbon-carbon double bond is present. The three most frequently encountered alkene groups are the vinyl, allyl, and isopropenyl groups, which may be named using suffixes: vinyloxy, allyloxy, methylenedioxy, allyl, and isopropenyl or with common names: methoxy, ethoxy, propoxy, and allyl.



vinyl group



allyl group



isopropenyl group

vinyl name: methoxy group      allyl name: allyloxy group      isopropenyl name: isopropenyl group

# Unsaturated Hydrocarbons Study Guide

**Jiben Roy**



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