

Chemical Reactions

Section 9.1 Reactions and Equations

In your textbook, read about evidence of chemical reactions.

For each statement, write *yes* if evidence of a chemical reaction is present. Write *no* if there is no evidence of a chemical reaction.

- | | |
|---------------|---|
| <u>yes</u> | 1. A tomato smells rotten. |
| <u>no</u> | 2. A drinking glass breaks into smaller pieces. |
| <u>no</u> | 3. A piece of ice melts. |
| <u>yes</u> | 4. Drain cleaner is mixed with water and the solution becomes warm. |
| <u>yes</u> | 5. Candle wax burns. |
| <u>yes/no</u> | 6. Molten candle wax solidifies. |
| <u>yes</u> | 7. Green leaves turn yellow and red as the seasons change. |
| <u>yes</u> | 8. Baking powder produces a gas that makes a cake rise. |

In your textbook, read about how to represent chemical reactions and how to balance chemical equations.

Use the terms below to complete the passage. Each term may be used once, more than once, or not at all.

arrow	plus sign	(s)	(l)
reactant	product	(g)	(aq)

The fuel for the space shuttle is hydrogen, which burns in oxygen to produce water vapor and energy. In this chemical reaction, hydrogen is a(n) **(9)** reactant, oxygen is a(n) **(10)** reactant, and water vapor is a(n) **(11)** product. In a chemical equation for this reaction, a(n) **(12)** arrow is used to separate hydrogen and oxygen from water vapor and energy. A(n) **(13)** plus sign is used to separate the symbols for hydrogen and oxygen. A(n) **(14)** (g) symbol is used to tell the state of hydrogen in the reaction, a(n) **(15)** (g) symbol is used for the state of oxygen, and a(n) **(16)** (g) symbol is used for the state of water vapor.

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Table of Contents Study Guide For Chemistry And Chemical Reactivity

1. Understanding the eBook Study Guide For Chemistry And Chemical Reactivity
 - The Rise of Digital Reading Study Guide For Chemistry And Chemical Reactivity
 - Advantages of eBooks Over Traditional Books
2. Identifying Study Guide For Chemistry And Chemical Reactivity
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Study Guide For Chemistry And Chemical Reactivity
 - User-Friendly Interface
4. Exploring eBook Recommendations from Study Guide For Chemistry And Chemical Reactivity

- Personalized Recommendations
 - Study Guide For Chemistry And Chemical Reactivity User Reviews and Ratings
 - Study Guide For Chemistry And Chemical Reactivity and Bestseller Lists
5. Accessing Study Guide For Chemistry And Chemical Reactivity Free and Paid eBooks
 - Study Guide For Chemistry And Chemical Reactivity Public Domain eBooks
 - Study Guide For Chemistry And Chemical Reactivity eBook Subscription Services
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 - Highlighting and Note-Taking Study Guide For Chemistry And Chemical Reactivity
 - Interactive Elements Study Guide For Chemistry And Chemical Reactivity
 8. Staying Engaged with Study Guide For Chemistry And Chemical Reactivity
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Study Guide For Chemistry And Chemical Reactivity
 9. Balancing eBooks and Physical Books Study Guide For Chemistry And Chemical Reactivity
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Study Guide For Chemistry And Chemical Reactivity
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Study Guide For Chemistry And Chemical Reactivity
 - Setting Reading Goals Study Guide For Chemistry And Chemical Reactivity
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Study Guide For Chemistry And Chemical Reactivity

- Fact-Checking eBook Content of Study Guide For Chemistry And Chemical Reactivity
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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