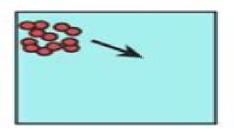
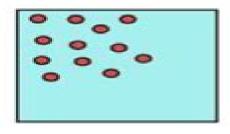
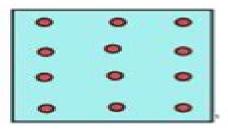
<u>Diffusion</u> is the random movement of particles from an area of high concentration to an area of low concentration until they are evenly spread out. When the molecules are evenly distributed throughout the space, they have reached "equilibrium".



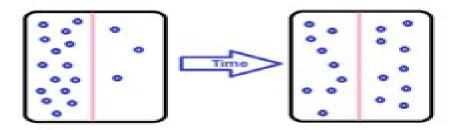




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A concentration gradient is a difference between concentrations in a space. In the boxes above, the particles in the first box show a concentration gradient, with a high concentration in the upper left corner and a low concentration in the lower right corner. Molecules will always move down the concentration gradient from an area of high concentration to an area of low concentration. Think of food coloring spreading out in water, or perfume moving through the air in a room.

After a time, the particles will reach equilibrium so that the particles are equally distributed and moving equally in all directions.



Unequal distribution

Equilibrium

^{1 &}lt;a href="http://www.biologycomer.com/resources/diffusion_red_JPG">http://www.biologycomer.com/resources/diffusion_red_JPG>

^{2 &}lt;a href="http://upload.wikimedia.org/wikipedia/commons/b/be/Simple Diffusion.png">http://upload.wikimedia.org/wikipedia/commons/b/be/Simple Diffusion.png>

Study Guide For Diffusion And Osmosis

Kevin T. Patton, Frank B. Bell, Terry Thompson, Peggie L. Williamson, Eric L Sun

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