

# 10.1 Periodic Trends



## 10.1 PERIODIC TRENDS: FROM CONCEPT TO PRACTICE

### Section 10.1: Introduction

In your notebook, read aloud the modern periodic table.

Copy the letter-headers in the first row on the left below from the periodic table to complete the table on the right.

_____
_____
_____
_____
_____

Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, either the letter of the matching item in Column B.

#### Column A

- \_\_\_\_\_ 10. It exists in the periodic table.
- \_\_\_\_\_ 11. It has no line periodic table.
- \_\_\_\_\_ 12. Group 1 elements.
- \_\_\_\_\_ 13. Elements that are strong oxidizing agents.
- \_\_\_\_\_ 14. Group 18 elements.

#### Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write each of the statements in terms of the statement in italics, change the underlined word or phrase to make it true.

- \_\_\_\_\_ 15. There are two representative elements of elements.
- \_\_\_\_\_ 16. Most elements have lines in the periodic table.
- \_\_\_\_\_ 17. Group 1 elements except for hydrogen are metals in the solid state.
- \_\_\_\_\_ 18. Group 18 elements are the active metal metals.
- \_\_\_\_\_ 19. Group 1 elements are highly reactive elements in the periodic table.
- \_\_\_\_\_ 20. Group 18 elements are very reactive elements found in the periodic table.
- \_\_\_\_\_ 21. Elements have properties of both metals and nonmetals.

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**K Payea**



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## **Table of Contents Study Guide For Stars Content Mastery Answers**

1. Understanding the eBook Study Guide For Stars Content Mastery Answers
  - The Rise of Digital Reading Study Guide For Stars Content Mastery Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Study Guide For Stars Content Mastery Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Study Guide For Stars Content Mastery Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Study Guide For Stars Content Mastery Answers
  - Personalized Recommendations
  - Study Guide For Stars Content Mastery Answers User Reviews and Ratings
  - Study Guide For Stars Content Mastery Answers and Bestseller Lists
5. Accessing Study Guide For Stars Content Mastery Answers Free and Paid eBooks
  - Study Guide For Stars Content Mastery Answers Public Domain eBooks
  - Study Guide For Stars Content Mastery Answers eBook Subscription Services
  - Study Guide For Stars Content Mastery Answers Budget-Friendly Options
6. Navigating Study Guide For Stars Content Mastery Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Study Guide For Stars Content Mastery Answers Compatibility with Devices
  - Study Guide For Stars Content Mastery Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Study Guide For Stars Content Mastery Answers
  - Highlighting and Note-Taking Study Guide For Stars Content Mastery Answers
  - Interactive Elements Study Guide For Stars Content Mastery Answers

8. Staying Engaged with Study Guide For Stars Content Mastery Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Study Guide For Stars Content Mastery Answers
9. Balancing eBooks and Physical Books Study Guide For Stars Content Mastery Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Study Guide For Stars Content Mastery Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Study Guide For Stars Content Mastery Answers
  - Setting Reading Goals Study Guide For Stars Content Mastery Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Study Guide For Stars Content Mastery Answers
  - Fact-Checking eBook Content of Study Guide For Stars Content Mastery Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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