

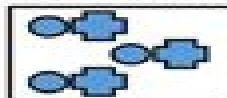
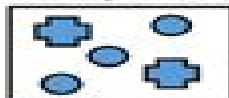
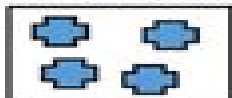
Unit 2: Chemistry
Unit Test Study Guide

Name _____

Unit 2 study Guide

8.P.1.1 Classify matter as elements, compounds, or mixtures based on how the atoms are packed together in arrangements.

- Vocab: atom, proton, neutron, electron, compound, element, mixture, homogeneous mixture, heterogeneous mixture
- Label the following as compound, mixture, or element:



8.P.1.2 Explain how the physical properties of elements and their reactivity have been used to produce the current model of the periodic table of elements.

- Vocab: periodic table, chemical symbol, chemical formula, metal, nonmetal, metalloid, conductivity, reactivity, atomic number, atomic mass, group/family, period, Noble gas family, Alkali family, Halogen family, valence electron, energy level
- How is the modern periodic table arranged?
- What is the difference between a group/family and a period?
- Which family on the periodic table is the least reactive?
- Which nonmetal family is the most reactive?
- What do elements in the same period have in common?
- What do elements in the same group have in common?
- What are the characteristics of metals? Where are they located?
- What are the characteristics of nonmetals? Where are they located?
- What are the characteristics of metalloids? Where are they located?
- What are valence electrons and how is their number determined?

8.P.1.3 Compare physical changes such as size, shape and state to chemical changes that are the result of a chemical reaction to include changes in temperature, color, formation of a gas precipitate.

- Vocab: acid, base, pH, chemical property, physical property, chemical reaction, physical change, solubility, solution, solute, solvent, catalyst, precipitate
- What are 5 signs that a chemical reaction has occurred?
- Is flammability a physical or chemical property?
- Is cutting paper a physical or chemical property?
- Is color a physical or chemical property?

Study Guide Logic Test Decathlon

Kathleen Armour



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