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Ch 5 Ions and Ionic Compounds Study Guide

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Identify the choice that best completes the statement or answers the question.

- 1. Which of the following electron configurations belongs to an element that is NOT chemically reactive?
 - a. $1x^22x^22p^3$ c. $1x^22x^22p^5$
 - b. 1s²2s²2p⁶3s²3p⁶
 d. 1s²2s²2p⁶3s²3p⁶4s¹
- In the compound sodium fluoride, NaF, the sodium atom loses one electron and the fluorine atom gains one electron to form ions that have electron configurations similar to
 - a. helium.

c. neon.

b. oxygen.

- d. calcium.
- In many compounds, atoms of main-group elements form ions so that the number of electrons in the outermost energy levels of each ion is
 - a. 2. b. 6.

- c. 8.d. 10.
- 4. The electron configuration of nitrogen is 1s²2s²2ρ³. How many more electrons does nitrogen need to have an electron configuration similar to neon?
 - a. 1b. 3

c. 5 d. 8

- ____ 5. Atoms of copper and iron
 - generally form stable bonds with transition elements.
 - b. have stable electron configurations.
 - c. tend to form cations.
 - d. tend to form anions.
- ____ 6. An anion
 - a. is an ion with a negative charge.
 - b. attracts ions with negative charges.
 - c. results when an alkaline-earth metal loses one of its two outermost electrons.
 - d. has more protons than electrons.
- When the octet rule is satisfied, the outermost _____ are filled
 - a. d and f orbitals

c. s and d orbitals

b. s and p orbitals

- d. a and p orbitals
- 8. The elements of the _____ group satisfy the octet rule without forming compounds.
 - a. halide

c. alkali metal

b. noble gas

- d. alkaline-earth metal
- Once an atom has full s and p orbitals in its outermost energy level.
 - a. it is highly reactive only with alkali metals.
 - b. it is highly reactive only with halogens.
 - it can be combined with most elements.
 - d. it has a stable octet and is unreactive.

Study Guide For Ions And Ionic Compounds

Susan McMurry

Study Guide For Ions And Ionic Compounds:

Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement Study Guide Central Hindu School Entrance Exam 2022 For Class 11 Arihant Experts, 2022-03-05 1 Central Hindu School Entrance Test is a complete test guide 2 Covers entire syllabus for class 11th 3 Topically divided into 5 sections to provide better understanding 4 Solved papers and Model papers are given for thorough practice The book CHS SET has been carefully designed to cater the needs of students of class 11th Encrypted with Chapterwise notes and previous years questions this book divides the entire syllabus into 5 major subjects Each chapter has been well explained in details to ease the understanding of the concepts Besides the theory part this book focuses on practice part as well with latest solved papers to get the insights of the exam pattern and two model papers for self assessment Housed with exam relevant content this study guide boosts the preparation level and raises the confidence of a student to score better in their exam TOC Model Solved Paper 2021 Arts Commerce Group Model Solved Papers 2021 Maths Bio Group Solved paper 2019 Art Commerce Group Solved Papers 2019 Maths Group Solved paper 2019 Bio Group English Hindi Mathematics Physics Chemistry Biology General Studies The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material

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