Activity A:	Get the Gizmo ready:	100
Activity A.	 Click Reset (2) and select Macro view. 	
Phase changes	 Set the Water temperature to 10 °C. 	2.0
	 Set the ice volume to 0 cm³. 	134

Question: How is temperature related to phase changes?

- 1. Predict: Based on your prior knowledge, predict the following:
 - A. At what temperature will water change from a liquid to a solid (freeze)?
 - B. At what temperature will water change from a solid to a liquid (melt)?
 - C. At what temperature will water change from a liquid to a gas (boil)?

0 100

0

- Investigate: Use the Gizmo to explore phase changes. Use the Add/remove heat energy slider to control
 the water temperature. Record your observations in your notes, then answer the questions below:
 - A. At what temperature does water freeze?
 - B. At what temperature does ice melt?
 - C. At what temperature does water boil?

This is the freezing point.

This is the melting point.

This is the boiling point.

Observe: Set up the Gizmo to observe freezing. What do you notice about the temperature while the water is in the process of freezing?

10

Remains unchanged

4. Explore: Use the Gizmo to investigate melting and boiling. Does the temperature change while either of these phase changes is occurring?

No it doesn't

5. <u>Interpret</u>: Select the GRAPH tab to see a graph of temperature vs. time. Click the "-" button until the whole graph is visible. What does the graph look like during a phase change?

A horizontal line

Extend your thinking: Why do you think the temperature does not change much during a phase change? If
possible, discuss your answer with your classmates and teacher.

Utilizing the energy to alter the nature of matter

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L Darling-Hammond

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- 1. Understanding the eBook Study Guide Phase Changes Answer Key
 - The Rise of Digital Reading Study Guide Phase Changes Answer Key
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Study Guide Phase Changes Answer Key
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Study Guide Phase Changes Answer Key
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Study Guide Phase Changes Answer Key
 - Personalized Recommendations
 - Study Guide Phase Changes Answer Key User Reviews and Ratings
 - Study Guide Phase Changes Answer Key and Bestseller Lists

- 5. Accessing Study Guide Phase Changes Answer Key Free and Paid eBooks
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 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - o Following Authors and Publishers Study Guide Phase Changes Answer Key
- 9. Balancing eBooks and Physical Books Study Guide Phase Changes Answer Key
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Study Guide Phase Changes Answer Key
- 10. Overcoming Reading Challenges
 - $\circ\,$ Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Study Guide Phase Changes Answer Key
 - Setting Reading Goals Study Guide Phase Changes Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Study Guide Phase Changes Answer Key
 - Fact-Checking eBook Content of Study Guide Phase Changes Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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