

Name _____ Date _____ Class _____

CHAPTER

15

STUDY GUIDE FOR CONTENT MASTERY

Solutions

Section 15.1 What are solutions?

In your textbook, read about the characteristics of solutions.

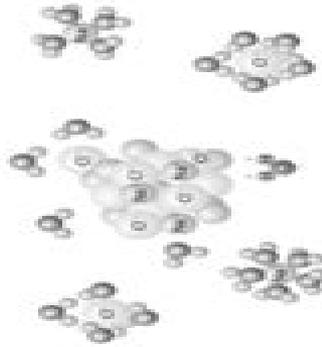
Use each of the terms below just once to complete the passage.

immiscible	liquid	soluble	solution
insoluble	miscible	solvent	solvent

Air is a(n) **(1)** solution of oxygen gas dissolved in nitrogen gas. The oxygen is an **(2)** solute and nitrogen is the **(3)** solvent. Because oxygen gas dissolves in a solvent, oxygen gas is an **(4)** soluble substance. A substance that does not dissolve is **(5)** insoluble. **(6)** Liquid solutions are the most common type of solutions. If one liquid is soluble in another liquid, such as acetic acid in water, the two liquids are **(7)** miscible. However, if one liquid is insoluble in another, the liquids are **(8)** immiscible.

Read about hydration in aqueous solutions in your textbook.

The diagram shows the hydration of solid sodium chloride to form an aqueous solution. Use the diagram to answer the following questions.



9. Hydration is hydration in which the solvent is water. What is hydration?
Solution is the process of surrounding solute particles with solvent particles to form a solution.

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CHAPTER

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STUDY GUIDE FOR CONTENT MASTERY

Section 15.1 continued

16. An sodium chloride dissolves in water. What happens to the sodium and chloride ions?
The sodium and chloride ions are separated and surrounded by the water molecules.

17. Explain the orientation of the water molecules around the sodium ions and chloride ions.
Because the sodium ion is positively charged, it attracts the negatively charged portion of the water molecule (the oxygen atom) and repels the positively charged portion of the water molecule (the hydrogen atoms). Because the chloride ion is negatively charged, it attracts the positively charged portion of the water molecule and repels the negatively charged portion.

18. How does the strength of the attraction between water molecules and sodium and chloride ions compare with the strength of the attraction between the sodium ions and chloride ions? How do you know?
The attraction between the water molecules and the sodium and chloride ions is greater than the attraction between the sodium and chloride ions. The greater strength of attraction between the water molecules and the ions is what causes the solvation process to occur.

19. List three ways that the rate of solvation may be increased.
stirring or shaking the solution, breaking the solute into smaller pieces, and heating the solvent

In your textbook, read about heat of solution, solubility, and factors that affect solubility.

For each statement below, write true or false.

- true** 20. The overall energy change that occurs when a solution forms is called the heat of solution.
false 21. Solubility is a measure of the maximum amount of solute that dissolves in a given amount of solvent at a specified temperature and pressure.
false 22. Solvation continues as long as the solvation rate is less than the crystallization rate.
true 23. In a saturated solution, solvation and crystallization are in equilibrium.
true 24. Additional water can be dissolved in an unsaturated solution.
true 25. The solubility of a gas dissolved in a liquid decreases as the temperature of the solution increases.

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Study Guide For Concept Mastery Answer Key

**Rich Weinfeld,Linda Barnes-
Robinson,Sue Jeweler,Betty Roffman
Shevitz**

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American Government/Politics Today 1999-2000 Schmidt, James Perkins, Shelley, 1998-12-16 **Memmler's The Human Body in Health and Disease, Enhanced Edition** Barbara Janson Cohen, Kerry L. Hull, 2020-08-03 Memmler's *The Human Body in Health and Disease* Fourteenth Edition is a textbook for introductory level allied health and nursing students who need a basic understanding of anatomy and physiology the interrelationships between structure and function and the effects of disease on body systems The features and content specifically meet the needs of those who may be starting their health career preparation with little or no science background The Fourteenth Edition has been revised and updated where needed to improve organization of the material and to reflect current scientific thought Each new print copy of Memmler's *The Human Body in Health Disease Enhanced 14th Edition* includes Navigate 2 Premier Access Includes The Body Visible a series of illustrations of the major body systems described in the text with labeled transparent overlays Retains an extensive art program with updated and new figures A main Glossary defines all the chapter's key terms and an additional Glossary of Word Parts is a reference tool that teaches basic medical and anatomic terminology and helps students learn to recognize unfamiliar terms Appendices include a variety of supplementary information that students will find useful as they work with the text including a new photographic Dissection Atlas answers to the Chapter Checkpoint questions Casepoint questions and Zooming In illustration questions that are found in every chapter Every chapter contains pedagogy that has been designed with the health professions and nursing student in mind 2019 694 pages

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