

# Introduction to Chemistry

## Section 1.1 The Stories of Two Chemicals

In your notebook, read aloud the entire paper.

Use each of the terms below just once to complete the passage.

stratosphere  
atmosphere

oxygen gas  
troposphere

ozone  
ultraviolet radiation

ozone hole

Earth's (1) \_\_\_\_\_ is made up of several layers. The air we breathe makes up the lowest level. This layer is called the (2) \_\_\_\_\_. The next layer up is called the (3) \_\_\_\_\_. This level contains a protective (4) \_\_\_\_\_ layer.

Ozone holes occur when (5) \_\_\_\_\_ is removed by ultraviolet radiation in the upper part of the atmosphere. The ozone forms a layer around Earth, which absorbs (6) \_\_\_\_\_. Without ozone, you are more likely to get a sunburn or possibly skin cancer. The thinning of the ozone layer, called the (7) \_\_\_\_\_, is a problem because without ozone all organisms on Earth are subject to more than one much radiation.

In your notebook, read aloud chlorofluorocarbons.

For each statement below, write true or false.

- \_\_\_\_\_ (A) CFC is another name for a chlorofluorocarbon.
- \_\_\_\_\_ (B) CFCs are made up of carbon, fluorine, and oxygen.
- \_\_\_\_\_ (C) All CFCs are synthetic chemicals.
- \_\_\_\_\_ (D) CFCs usually react easily with other chemicals.
- \_\_\_\_\_ (E) CFCs were developed as replacement for toxic refrigerants.

# Study Guide For Content Mastery Answer Key Chemistry Chapter 1

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RN, MSN, CCRN**



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