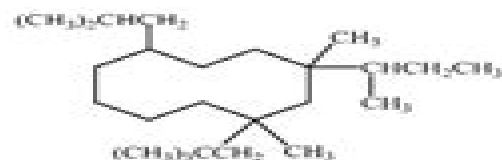


INSTRUCTIONS (Read carefully): Carefully separate all sheets. This examination has 2 pages of problems, 1 double-sided answer sheet, a periodic table, and 1 blank sheet for scratch work. Read each question carefully. For 2 **bonus points**, correctly complete as requested on *both sides* of the answer sheet: print your name, sign your name, give the portion of your FSU e-mail address that appears before the @ (this is the same as your BlackBoard login user name), give your row letter and seat number, and give your Discussion Section number. Put your answers in the spaces provided on the answer sheet. *All sheets will be collected; only answers on the answer sheets will be graded.* There is a total of 100 points plus 2 bonus points. Before the next lecture, go over problems about which you were not confident on this examination.

REGRADE POLICY: If you have a question about the number of points that you earned on your graded exam answer sheet, *do not make changes or additions to your returned, graded answer sheet.* Write on a *separate sheet of paper* what items you believe that you were not given sufficient credit and the reasons why you think so. Attach it to your original answer sheet and give it to Dr. Hilinski after a lecture or during an office hour, preferably within 2 weeks after it is returned. Be sure to make and keep a photocopy for your records.

- [9 points] For the compound given to the right, provide a name.
- [12 points] For each of the following, consider the compound given in *problem 1*.



- How many methylene hydrogens are there?
- How many tertiary Cs are there?
- Give the molecular formula of this compound.
- For the *number of Cs* in this compound, what is the maximum number of Hs? Do not consider only the specific structural formula given in *problem 1*; do consider the general case for the number of Cs that you gave in *part 2c*.
- How many tertiary Hs are there?
- How many primary alkyl groups exist as substituents?

- [7 points] Provide a structural formula for 5-ethyl-2,3-dimethyl-8-(1-methylcyclopentyl)nonane. *For this problem, you must give a complete structural formula showing all atoms and all bonding valence electrons.*
- [22 points] Consider the following.



- One of the 5 resonance structures (Lewis structures) that contribute to the resonance hybrid of the conjugate base of phenol is represented above as one of the products in the given chemical equation. Draw the additional 4 resonance structures (Lewis structures) that contribute to the resonance hybrid of the conjugate base of phenol.

Note that if you incorrectly duplicate resonance structures, you will lose credit for that resonance structure. If you give more than 4 resonance structures, points will be deducted.

For each of the four Lewis structures (resonance structures) that you give, you must show all atoms, all bonding valence electrons, all unshared valence electrons, and each nonzero formal charge.

- The equilibrium shown above lies to the right. This, of course, means that phenol is more acidic than cyclohexanol. Using structural considerations, provide a clear, concise explanation (but in as much detail as possible) why this equilibrium lies to the right. **NOTE** that a correct answer to this part is *not* that "the pK_a of phenol is less than the pK_a of cyclohexanol" because this is merely a restatement of the fact that phenol is more acidic than cyclohexanol.

- [10 points] Draw structural formulas for 5 (*and only 5*) constitutional isomers that *have one (and only one) ring* and the molecular formula C_4H_5FO . Yes, each of the constitutional isomers that you give must be *have one (and only one) ring*. Your structural formulas must have no nonzero formal charges and the most typical arrangements of bonding and nonbonding valence electrons. *Note that if you give the same isomer more than once, points will be deducted for that isomer. If you give more than 5 constitutional isomers, points will be deducted. For this problem, you must give complete structural formulas showing all atoms and bonding valence electrons. Do not give condensed structural formulas or bond-line formulas.*
- [4 points] A condensed structural formula for allene is shown to the right. Using lines, wedges, and dashed wedges for bonding electrons, give a 3-D stereoisomer for allene that is a representation of the true shape of this molecule. For your structure, label clearly each C with its hybridization (sp^3 , sp^2 , or sp).



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Marye Anne Fox, James K. Whitesell



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Organic Chemistry, 4e Student Solution Manual and Study Guide David R. Klein, 2021-01-07 Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems With Organic Chemistry Student Solution Manual and Study Guide 4th Edition students can learn to become proficient at approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry Organic Chemistry Study Guide with Solutions Manual Neil E. Schore, 2007 The guide includes chapter introductions that highlight new material chapter outlines detailed comments for each chapter section a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer A *Self-study Guide to the Principles of Organic Chemistry* Jiben Roy, 2013 A Self Study Guide to the Principles of Organic Chemistry Key Concepts Reaction Mechanisms and Practice Questions for the Beginner will help students new to organic chemistry grasp the key concepts of the subject quickly and easily as well as build a strong foundation for future study Starting with the definition of atom the author explains molecules electronic configuration bonding hydrocarbons polar reaction mechanisms stereochemistry reaction varieties organic spectroscopy aromaticity and aromatic reactions biomolecules organic polymers and a synthetic approach to organic compounds The over one hundred diagrams and charts contained in this volume will help students visualize the structures and bonds as they read the text and make the logic of organic chemistry clear and easily understood Each chapter ends with a list of frequently asked questions and answers followed by additional practice problems Answers are included in the Appendix *Study Guide and Solutions Manual for Essential Organic Chemistry* Paula Yurkanis Bruice, 2015-03-04 This Study Guide Solution Manual contains learning objectives chapter summaries and outlines as well as examples self tests and concept questions as well as complete step by step solutions to selected problems **Organic Chemistry, 5e Student Study Guide and Solutions Manual** David R. Klein, Laurie S. Starkey, 2025-03-18 Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems With Organic Chemistry Student Study Guide and Solutions Manual 5th Edition students can learn to become proficient at approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry Organic Chemistry Study Guide Robert J. Ouellette, J. David Rawn, 2014-11-04 Organic Chemistry Study Guide Key Concepts Problems and Solutions features hundreds of problems from the companion book Organic Chemistry and includes solutions for every problem Key concept summaries reinforce critical material from the primary book and enhance mastery of this complex subject Organic chemistry is a constantly evolving field that has great relevance for all scientists not just chemists For chemical engineers understanding the properties of organic molecules and how reactions occur is critically important to understanding the processes in an industrial plant For biologists and health professionals it is essential because nearly all of biochemistry

springs from organic chemistry. Additionally, all scientists can benefit from improved critical thinking and problem-solving skills that are developed from the study of organic chemistry. Organic chemistry, like any skill, is best learned by doing. It is difficult to learn by rote memorization, and true understanding comes only from concentrated reading and working as many problems as possible. In fact, problem sets are the best way to ensure that concepts are not only well understood but can also be applied to real-world problems in the workplace. Helps readers learn to categorize, analyze, and solve organic chemistry problems at all levels of difficulty. Hundreds of fully worked practice problems, all with solutions. Key concept summaries for every chapter reinforce core content from the companion book.

Organic Chemistry, Student Study Guide and Solutions Manual David R. Klein, 2017-01-04. This is the Student Study Guide and Solutions Manual to accompany Organic Chemistry 3e. Organic Chemistry 3rd Edition is not merely a compilation of principles but rather it is a disciplined method of thought and analysis. Success in organic chemistry requires mastery in two core aspects: fundamental concepts and the skills needed to apply those concepts and solve problems. Readers must learn to become proficient at approaching new situations methodically based on a repertoire of skills. These skills are vital for successful problem solving in organic chemistry. Existing textbooks provide extensive coverage of the principles, but there is far less emphasis on the skills needed to actually solve problems.

Organic Chemistry, 13e Student Study Guide and Solutions Manual T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2022-05-03. Organic Chemistry Student Study Guide and Solutions Manual 13th Edition offers the full solutions for select exercises from the text.

Study Guide, Answer Book to Accompany Organic Chemistry Alan S. Wingrove, Robert L. Caret, 1981.

Organic Chemistry, 12e Binder Ready Version Study Guide & Student Solutions Manual T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2016-04-11. This is the Student Study Guide Solutions Manual to accompany Organic Chemistry 12th Edition. The 12th edition of Organic Chemistry continues Solomons Fryhle Snyder's tradition of excellence in teaching and preparing students for success in the organic classroom and beyond. A central theme of the authors' approach to organic chemistry is to emphasize the relationship between structure and reactivity. To accomplish this, the content is organized in a way that combines the most useful features of a functional group approach with one largely based on reaction mechanisms. The authors' philosophy is to emphasize mechanisms and their common aspects as often as possible and at the same time use the unifying features of functional groups as the basis for most chapters. The structural aspects of the authors' approach show students what organic chemistry is. Mechanistic aspects of their approach show students how it works. And wherever an opportunity arises, the authors show students what it does in living systems and the physical world around us.

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