

Name _____ Date _____ Class _____

CHAPTER 15 STUDY GUIDE FOR CONTENT MASTERY

Solutions

Section 15.1 What are solutions?

In your textbook, read about the characteristics of solutions.

Use each of the terms below just once to complete the passage.

immiscible	liquid	soluble	solution
insoluble	miscible	solvent	solvent

Air is air! (1) solution of oxygen gas dissolved in nitrogen gas. The oxygen is air is the (2) solvent and nitrogen is the (3) solvent. Because oxygen gas dissolves in a solvent, oxygen gas is air is (4) soluble substances. A substance that does not dissolve is (5) insoluble. (6) liquid solutions are the most common type of solutions. If one liquid is soluble in another liquid, such as acetic acid in water, the two liquids are (7) miscible. However, if one liquid is insoluble in another, the liquids are (8) immiscible.

Read about solvation in aqueous solutions in your textbook.

The diagram shows the hydration of solid sodium chloride to form an aqueous solution. Use the diagram to answer the following questions.



8. Hydration is solvation in which the solvent is water. What is solvation?

Solvation is the process of surrounding solute particles with solvent particles to form a solution.

Study Guide for Content Mastery

Chemistry: Matter and Change • Chapter 15

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CHAPTER 15 STUDY GUIDE FOR CONTENT MASTERY

Section 15.1 continued

10. As sodium chloride dissolves in water, what happens to the sodium and chloride ions?

The sodium and chloride ions are separated and surrounded by the water molecules.

11. Explain the attraction of the water molecules around the sodium ions and chloride ions.

Because the sodium ion is positively charged, it attracts the negatively charged portion of the water molecule (the oxygen atom) and repels the positively charged portion of the water molecule (the hydrogen atoms). Because the chloride ion is negatively charged, it attracts the positively charged portion of the water molecule and repels the negatively charged portion.

12. How does the strength of the attraction between water molecules and sodium and chloride ions compare with the strength of the attraction between the sodium ions and chloride ions? How do you know?

The attraction between the water molecules and the sodium and chloride ions is greater than the attraction between the sodium and chloride ions. The greater strength of attraction between the water molecules and the ions is what causes the solvation process to occur.

13. List three ways that the rate of solvation may be increased.

stirring or shaking the solution, breaking the solute into smaller pieces, and heating the solvent

In your textbook, read about heat of solution, solubility, and factors that affect solubility.

For each statement below, write true or false.

True _____ 14. The overall energy change that occurs when a solution forms is called the heat of solution.

False _____ 15. Solubility is a measure of the maximum amount of solute that dissolves in a given amount of solvent at a specified temperature and pressure.

False _____ 16. Solvation continues as long as the solvation rate is less than the crystallization rate.

True _____ 17. In a saturated solution, solvation and crystallization are in equilibrium.

True _____ 18. Additional solute can be dissolved in an unsaturated solution.

True _____ 19. The solubility of a gas dissolved in a liquid decreases as the temperature of the solution increases.

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Study Guide for Content Mastery

Study Guide For Content Master Answer Key

Donna D. Ignatavicius, Cherie R. Rebar

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Study Guide & Laboratory Manual for Physical Examination & Health Assessment E-Book Carolyn Jarvis, Ann L. Eckhardt, 2023-01-02 Both a comprehensive lab manual and a practical workbook the Study Guide Laboratory Manual for Physical Examination Health Assessment 9th Edition gives you the tools you need to master physical examination and health assessment skills in the lab and in clinical practice Corresponding to the bestselling Jarvis textbook this guide features terminology reviews application activities clinical judgment questions regional write up sheets and narrative summary forms with answers to study questions at the back of the book to facilitate both learning and review The 9th edition has been thoroughly updated with a fresh focus on the Next Generation NCLEX NGN with case studies featuring new NGN question formats to prepare you not only for the skills laboratory but for success on the NCLEX and in interprofessional collaborative practice Authoritative review and guidance for laboratory experiences personally written by the textbook authors provide a seamlessly integrated study and clinical experience Consistent format includes Terminology Review Study Guide and Clinical Judgment Questions in each chapter Physical examination forms familiarize you with what you will encounter in clinical practice and offer practice in documenting the patient history and examination findings The only full color illustrated lab manual available for a nursing health assessment textbook with anatomy exercises that align with the main text NEW Clinical judgment exercises equip you for success on the Next Generation NCLEX NGN including questions with an increased focus on clinical judgment robust single episode case studies that employ the latest NGN question types and unfolding case studies which reflect the language of the NCSBN Clinical Judgment Measurement Model NEW Chapter 33 includes unfolding case studies for the NGN that present opportunities for you to practice prioritizing decision making and using clinical judgment skills NEW Increased emphasis on activities focused on higher cognitive levels Applying and above UPDATED Critical Thinking Exercises offer suggested readings based on your participation in the skills lab and discussions with your instructor UPDATED Content corresponds to the 9th edition of the Jarvis textbook and incorporates the latest research and evidence based practice

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Streamlined ID Miriam B. Larson, Barbara B. Lockee, 2013-08-22 Streamlined ID A Practical Guide to Instructional Design presents a focused and generalizable approach to instructional design and development one that addresses the needs of ID novices as well as practitioners in a variety of career environments Emphasizing the essentials and big ideas of ID Streamlined ID presents a new perspective one that aims to produce instruction that is sustainable optimized appropriately redundant and targeted at continuous improvement The book features an enhanced version of the classic ADDIE model Analysis Design Development Implementation and Evaluation that emphasizes the iterative nature of design and the role of evaluation throughout the design development process It clearly lays out a systematic approach that emphasizes the use of research based theories while acknowledging the need to customize the process to address a variety of pedagogical approaches Instructivist Constructivist and Connectivist The book opens with an overview of the basics of ID and each subsequent chapter describes major activities in the ID process with step by step instructions and tips for streamlining the process Numerous job aids serve to maximize the efficiency and effectiveness of your design efforts Each chapter highlights key concepts and provides additional exercises and assignments based on the work of Benjamin Bloom Streamlined ID is an ideal reference guide for optimizing professional practice

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