Name	Date

Chapter 12 – Stoichiometry Study Guide

The following example problems exhibit the types of calculations you will be expected to perform on tomorrow's test.

Describe a reaction in terms of particles

Ex.) Describe the following reaction in terms of molecules

Note: sometimes you will need to include other descriptors such as atoms and formula units*

Describe a reaction in terms of moles
 Ex.) Describe the following reaction in terms of moles

How do you determine the mass of the products for a balanced reaction?
 Ex.) What is the mass of the product formed in the following equation?

$$H_2 + CI_2 \rightarrow 2HCI$$

 How do you determine the volume of the reactants or products of a balanced reaction?

Ex.) What is the volume of the products and the reactants of the following equation?

$$2H_2 + O_2 \rightarrow 2H_2O$$

 How do you show how a balanced reaction follows the Law of Conservation of Mass by showing that the mass on both sides of a reaction are equal?

Ex.) Show how the mass of the reactants equals the mass of the products for the following equation.

SJ Ball

STOICHIOMETRY AND PROCESS CALCULATIONS K. V. NARAYANAN.B. LAKSHMIKUTTY.2006-01-01 This textbook is designed for undergraduate courses in chemical engineering and related disciplines such as biotechnology polymer technology petrochemical engineering electrochemical engineering environmental engineering safety engineering and industrial chemistry. The chief objective of this text is to prepare students to make analysis of chemical processes through calculations and also to develop in them systematic problem solving skills. The students are introduced not only to the application of law of combining proportions to chemical reactions as the word stoichiometry implies but also to formulating and solving material and energy balances in processes with and without chemical reactions The book presents the fundamentals of chemical engineering operations and processes in an accessible style to help the students gain a thorough understanding of chemical process calculations It also covers in detail the background materials such as units and conversions dimensional analysis and dimensionless groups property estimation P V T behaviour of fluids vapour pressure and phase equilibrium relationships humidity and saturation With the help of examples the book explains the construction and use of reference substance plots equilibrium diagrams psychrometric charts steam tables and enthalpy composition diagrams It also elaborates on thermophysics and thermochemistry to acquaint the students with the thermodynamic principles of energy balance calculations Key Features SI units are used throughout the book Presents a thorough introduction to basic chemical engineering principles Provides many worked out examples and exercise problems with answers Objective type questions included at the end of the book serve as useful review material and also assist the students in preparing for competitive examinations such as GATE **Resources in Education** ,2000-10 Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 A-level Chemistry Complete Guide (Concise) (Yellowreef) Thomas Bond, Chris Hughes, 2013-11-04 covers latest MOE syllabus and beyond enable accurate complete and independent self education holistic question answering techniques examples include mark schemes and exam reports the only quide currently that teaches Planning Questions available only in print edition and complete edition eBook advanced trade book Complete edition and concise edition eBook available Master Dentistry Volume 3 Oral Biology Barry Berkovitz, Bernard J. Moxham, Roger W. A. Linden, Alastair J. Sloan, 2010-09-17 Master Dentistry Oral Biology is designed as a revision guide for dental students and offers the curriculum essentials in an easy to digest format Each section is fully illustrated throughout and is supported by extensive self assessment questions which allow the reader to assess their own knowledge of the topic and perfect their exam techniques New to the Master Dentistry series this volume addresses oral anatomy and its related histology physiology and biochemistry and is perfect for both undergraduate students and post graduates preparing for the MJDF Quick reference revision aid for dental students ideal for exam preparation Covers the essentials of the subject to a level that is expected with the GDC s curriculum outlined in the First Five Years document Each chapter provides a brief

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