Type 1 (Ionic) Bonding Worksheet

For each pair of elements below draw electron dot structures showing the valence electrons in each atom. Then draw arrows to show where the valence electrons will go during a chemical reaction. Write the name of each type I ionic compound. Finally, fill in the n table below each reaction. Refer to the sample shown.

		• •				4		8 <u>9</u> 28
	•	\mathbf{F}	:		Li	=	\mathbf{F}	:
Li								

Reactions- draw a picture showing each reaction & name the ionic compound	Atoms	Valence electrons	Electron transfer from/to each atom	lons formed in the product
Li + F => LiF Name the ionic compound – lithium fluoride	u	1	Li loses 1 e-	n.
	F	7	F gains 1 e-	E t-
Ca + O => CaO Name the ionic compound – calcium oxide	Ca	2	Ca loses 2 e-	Ca 2
	0	6	O gains 2 e-	0.3
3) Be + F => BeF2 Name the ionic compound -				
Name the ionic compound -				
3) Mg + S => MgS				
Name the ionic compound				-
4) K+F=>KF				
Name the ionic compound				

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Dr. O. P. Agarwal

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Student Solutions Manual Darrell Ebbing, 2005-03-17 Provides solutions to odd numbered Practice Problems General Problems and Cumulative Skills Problems plus answers to Review Questions **Ebbing** Darrell D. Ebbing, 1993 5 Chemistry: Practice Papers for SQA Exams Barry McBride, 2016-10-03 Exam Board SQA Level National 5 Subject Chemistry First Teaching September 2013 First Exam Summer 2014 Practise for your SQA exams with three specially commissioned Hodder Gibson Practice Exam Papers with fully worked answers Practise with model papers written and checked by experienced markers and examiners Worked answers show how solutions are arrived at and where marks are gained Get extra advice with study skills guidance sections Avoid common mistakes with examiner tips A revision grid allows students to Essential SQA Exam Practice: National 5 Chemistry Questions and Papers Barry revise by topic McBride, 2019-12-02 Exam board SQA Level National 5 Subject Chemistry First teaching September 2017 First exam Summer 2018 Practice makes permanent Feel confident and prepared for the SQA National 5 Chemistry exam with this two in one book containing practice questions for every question type and topic plus two full practice papers Choose to revise by question type or topic A simple grid enables you to pick particular question styles or course areas that you want to focus on with answers provided at the back of the book Understand what the examiner is looking for Clear guidance on how to answer each question type is followed by plenty of questions so you can put the advice into practice building essential exam skills Remember more in your exam Repeated and extended practice will give you a secure knowledge of the key areas of the course chemical changes and structure nature s chemistry chemistry in society Familiarise yourself with the exam paper Both practice papers mirror the language and layout of the real SOA papers complete them in timed exam style conditions to increase your confidence before the exams Find out how to achieve a better grade Answers to the practice papers have commentaries for each question with tips on writing successful answers and avoiding common mistakes Fully up to date with SQA's requirements. The questions mark schemes and quidance in this practice book match the requirements of the revised SQA National 5 Chemistry specification for examination from 2018 onwards **AP Chemistry For Dummies** Peter J. Mikulecky, Michelle Rose Gilman, Kate Brutlag, 2008-11-13 A practical and hands on guide for learning the practical science of AP chemistry and preparing for the AP chem exam Gearing up for the AP Chemistry exam AP Chemistry For Dummies is packed with all the resources and help you need to do your very best Focused on the chemistry concepts and problems the College Board wants you to know this AP Chemistry study guide gives you winning test taking tips multiple choice strategies and topic guidelines as well as great advice on optimizing your study time and hitting the top of your game on test day This user friendly guide helps you prepare without perspiration by developing a pre test plan organizing your study time and getting the most out or your AP course You ll get help understanding atomic structure and bonding grasping atomic geometry understanding how colliding particles produce states and so much more To provide students with hands on

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hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement

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