

NAME: _____ EQUILIBRIUM PRACTICE

Equilibrium Practice

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1. The reaction below has reached a state of equilibrium. Write the equilibrium constant expression for this reaction.
2. Consider the following chemical equation: $2\text{H}_2\text{O}(g) \rightleftharpoons 2\text{H}_2(g) + \text{O}_2(g)$. In equilibrium, the concentration of H_2 is 0.02 M, the concentration of H_2O is 0.01 M, and the concentration of O_2 is 0.01 M. Calculate the equilibrium constant, K .
3. At a certain pressure, 10 g of each of the gases, chlorine and hydrogen, were placed in a 100 L container. Calculate the equilibrium constant, K , for the following equation: $\text{H}_2 + \text{Cl}_2 \rightleftharpoons 2\text{HCl}$. At equilibrium, the concentration of each gas was 0.01 M.
4. A 100 mL flask is filled with a gas mixture at a pressure of 1 atm. The flask is sealed and the gas mixture is allowed to reach equilibrium. Calculate the equilibrium constant, K , for the reaction: $\text{H}_2 + \text{Cl}_2 \rightleftharpoons 2\text{HCl}$.
5. Consider the following chemical equation: $\text{PCl}_5(g) \rightleftharpoons \text{PCl}_3(g) + \text{Cl}_2(g)$. At the initial concentration of PCl_5 is 1.0 M, what is the equilibrium concentration of Cl_2 if the concentration of PCl_3 is 0.5 M?

Skill Practice 45 Equilibrium Practice Answers

Robert Courtney Smith



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Intimations Pursued Andrew Sullivan, 2016-06-07 In this book Andrew Sullivan examines Oakeshott's transition from his original emphasis on philosophy as providing what was ultimately satisfactory in experience to his later emphasis on practical life This satisfaction is best achieved by a fusion of the modes of poetry and practice leading the author to examine Oakeshott's view of religious life as the consummation of practice in its most poetic incarnation The book also examines how the conception of practice is applied in Oakeshott's political writings focusing on the notion of civil association

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