

Section 6-1

Concept Review

Section: Simple Ions

Complete each statement below by choosing a term from the following list. Terms may be used more than once.

10

11

ion

octet

12

13

anion

cation

1. An atom or group of atoms that has a positive or negative electric charge because it has lost or gained electrons is a(n) ion.
2. The sodium ion has 11 protons, 10 neutrons, and 12 electrons.
3. In most chemical reactions, atoms tend to match the outer electron configuration of the noble gases. This is called the octet rule.

Complete each statement below by writing the correct term or phrase.

4. An anion is an ion with a negative charge.
5. Valence electrons occupy the outermost energy level of an atom.
6. The chloride ion and the chlorine atom have different electron configurations.
7. A cation is an ion with a positive charge.
8. Many stable ions have an electron configuration of a octet.

Answer the following questions in the space provided.

9. How do the outer-shell electron configurations for ions of Group 1, Group 2, and Group 15, Group 16, and Group 17 elements compare with those of the noble gases?

noble gases are stable while everything else is missing some

Skills Worksheet Section Simple Ions

Nancy Steineke

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Holt Chemistry Holt Rinehart & Winston, 2003-01-24 **DBT Skills Training Manual** , Te HS&T 2007 Shrt Crs M
Holt Rinehart & Winston, 2007 **Teacher's Guide to Effective Sentence Writing** Bruce Saddler, 2012-10-01 This practical book provides explicit instructions for teaching sentence level skills to students who have difficulties in this area. The author explains the key role of sentence combining in the writing process and presents effective techniques for instruction and assessment. Numerous sample lessons practice activities planning tips and grammatical pointers make it easy for teachers to incorporate sentence combining and construction into the writing curriculum at all grade levels 2-12. Accessible and engaging the book helps teachers and students experiment with different ways to arrange thoughts and produce meaningful written work.

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