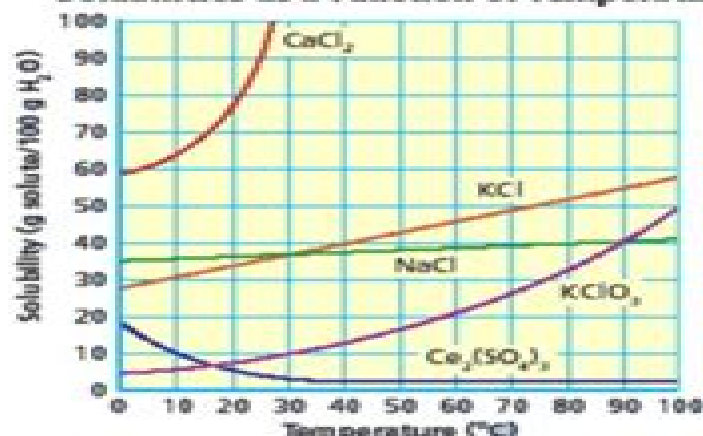


Name: _____

Date: _____ Period: _____

Solubility Curve Worksheet 2

Directions: Use the graph below to answer the following questions.

Solubilities as a Function of Temperature**Types of Solutions**

- > Points that collectively make up the solubility curve (points **ON** the curve) represent _____ solutions.
- > Points that are **BELOW** the curve represent _____ solutions.
- > Points **ABOVE** the solubility curve represent _____ solutions, and the **difference** between the point above the curve and on the curve represents the amount of solute which will precipitate out.
- > Say if the resulting solutions would be saturated, supersaturated or unsaturated.
 - o 60 g of KCl at 70°C _____
 - o 10 g of KClO₃ at 60°C _____
 - o 70 g of CaCl₂ at 20°C _____

1. a. What is the solubility of calcium chloride (CaCl₂) at 5°C? _____
- b. What is the solubility of calcium chloride at 25°C? _____
- c. What is happening to the solubility and temperature for CaCl₂? Solubility is _____ as temperature is _____.
2. a. What is the solubility of cerium sulfate - Ce₂(SO₄)₃ at 10°C? _____
- b. What is the solubility of cerium sulfate at 50°C? _____
- c. What is happening to the solubility and temperature for Ce₂(SO₄)₃? Solubility is _____ as temperature is _____.
3. Which substance on the graph is **least** soluble at 10°C? _____
4. Which substance on the graph shows the **least** change in solubility from 0°C to 100°C? _____
5. Which solution is more concentrated:
 - a) At 10°C, a saturated solution of NaCl or a saturated solution of CaCl₂ (circle one)
 - b) At 55°C, a saturated solution of KClO₃ or a saturated solution of Ce₂(SO₄)₃ (circle one)
6. What is the mass of cerium sulfate that will dissolve in 50 g of water at 10 degrees Celsius? _____
7. What is the mass of potassium chloride that will dissolve in 250 g of water at 70 degrees Celsius? _____
8. At 90 degrees Celsius, 10 g of potassium chlorate is dissolved in 100 g of water. Is this solution saturated, unsaturated, or supersaturated? How do you know? _____
9. A saturated solution of potassium chlorate is dissolved in 100g of water. If the saturated solution is cooled from 90 degrees Celsius to 60 degrees Celsius, how many grams would crystallize out? _____

Solubility Curve 2 Answers

Maxwell Gelender

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

Solubility Curve 2 Answers:

Regents Chemistry--Physical Setting Power Pack Revised Edition Barron's Educational Series, Albert S. Tarendash, 2021-01-05 Barron's two book Regents Chemistry Power Pack provides comprehensive review actual administered exams and practice questions to help students prepare for the Chemistry Regents exam This edition includes Regents Exams and Answers Chemistry Eight actual administered Regents Chemistry exams so students can get familiar with the test Thorough explanations for all answers Self analysis charts to help identify strengths and weaknesses Test taking techniques and strategies A detailed outline of all major topics tested on this exam A glossary of important terms to know for test day Let's Review Regents Chemistry Extensive review of all topics on the test Extra practice questions with answers A detailed introduction to the Regents Chemistry course and exam One actual recently released Regents Chemistry exam with an answer key ACT Total Prep 2025: Includes 2,000+ Practice Questions + 6 Practice Tests Kaplan Test Prep, 2024-06-04 ACT Total Prep 2025 Kaplan's biggest ACT prep book has the most content review efficient strategies and realistic practice to help you score higher We have everything you need in one big book plus a full year of access to online resources including more practice tests a bigger Qbank than ever 500 questions and video lessons to help you master each section of the ACT Publisher's description *E3 Chemistry Review Book - 2018 Home Edition (Answer Key Included)* Effiong Eyo, 2017-10-20 With Answer Key to All Questions Chemistry students and homeschoolers Go beyond just passing Enhance your understanding of chemistry and get higher marks on homework quizzes tests and the regents exam with E3 Chemistry Review Book 2018 With E3 Chemistry Review Book students will get clean clear engaging exciting and easy to understand high school chemistry concepts with emphasis on New York State Regents Chemistry the Physical Setting Easy to read format to help students easily remember key and must know chemistry materials Several example problems with solutions to study and follow Several practice multiple choice and short answer questions at the end of each lesson to test understanding of the materials 12 topics of Regents question sets and 3 most recent Regents exams to practice and prep for any Regents Exam This is the Home Edition of the book Also available in School Edition ISBN 978 197836229 The Home Edition contains an answer key section Teachers who want to recommend our Review Book to their students should recommend the Home Edition Students and parents whose school is not using the Review Book as instructional material as well as homeschoolers should buy the Home Edition The School Edition does not have answer key in the book A separate answer key booklet is provided to teachers with a class order of the book Whether you are using the school or Home Edition our E3 Chemistry Review Book makes a great supplemental instructional and test prep resource that can be used from the beginning to the end of the school year PLEASE NOTE Although reading contents in both the school and home editions are identical there are slight differences in question numbers choices and pages between the two editions Students whose school is using the Review Book as instructional material SHOULD NOT buy the Home Edition Also available in paperback print

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Review Text in Chemistry Maxwell Gelender,1959

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