compounds	Exceptions	soluble compounds	Exceptions
Group 1 ions (Li+,Na+, K+,)	70	Carbonate (CO ₃ ² -)	When combined with group 1 ions or ammonium (NH ₄ +)
Ammonium (NH4+)	-	Chromate (CrO ₄ ² -)	When combined with group 1 ions, Ca ²⁺ , Mg2+, or ammonium (NH ₄ +)
Nitrate (NO ₃ -)		Phosphate (PO ₄ ³⁻)	When combined with group 1 ions or ammonium (NH ₄ *)
Acetate (C ₂ H ₃ O ₂ - or CH ₃ COO-)		Sulfide (S ²⁻)	When combined with group 1 ions or ammonium (NH ₄ *)
Hydrogen Carbonate (HCO3 ⁻)		Hydroxide (OH·)	When combined with group 1 ions, Ca ^{2*} , Ba ^{2*} , Sr ^{2*} or ammonium (NH ₄ *)
Chlorate (ClO ₃ -)		Sulfates (SO ₄ ² ·)	When combined with Ag+, Ca ²⁺ , Sr ²⁺ , Ba ²⁺ , and Pb ²⁺
Perchlorate (ClO ₄ -)	20	Halides (Cl-, Br-, I-)	When combined with Ag*, Pb ² *, and Hg ₂ ² *
vailable for the reaction dosimetry issues. For ource for spinal cancer atalyst solution. This so result in a homogeno roperly mix with the ca	n. It may affect example, in B treatment wa dution is then us product. T talyst solution act constant,	t distribution in the ra- enega et. al. [34] a pho s developed by mixing added to an epoxy res his won't be achievabl (water based). Ksp, is the equilibriu	amount of radioisotope diation source resulting esphorus-32 radioactive the radioisotope with a sin. Both solutions need le if the isotope doesn't m constant for a solid e level at which a solute
issolve in solution. Equ	ation 17 show	s the correlation.	
	$aA_{(e)} \leftrightarrow$	$bB_{(aa)} + cC_{(aa)}$	(15)

Ions that form

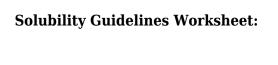
Exceptions

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Anthony S. Fauci



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