

Net Ionic Equations

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- Net ionic equations show only the compounds that undergo a chemical change in the reaction (look for changes in state).
- Ones that do not participate are known as spectator ions.
 These are canceled to form the

net ionic equation.

Ammonium sulfide reacts with cadmium (II) nitrate

NIE -
$$Cd^{2*}_{(aq)}$$
 + $S^{2*}_{(aq)}$ $\rightarrow CdS_{(s)}$

Solubility Rules And Net Ionic Equations

AN Whitehead

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