



The following data refer to an experiment in which a measured mass of solid is added to 10.0 g of 20 °C water. The mooture is crieved and allowed to sit for 5 hours. Tim separate trials are conducted for the experiment.

Tirial Number	Mass of reduce added (grams)	Mans of solute dissolved (grams)	**T
365	1.0	3.0	2-
2	2.0	2.0	284
35	5.0	3.0	2-
4	6.0	3.6	
5	5.0	3.6	1.5
6.	6.0	3.6	
77	2.00	8.6	17
	8.0	3,6	**
9	9.0	3.6	La transport
10	10.0	5.6	0-04



Dependent variable marrie who make the Absorb seven all temporal. Independent variable market the time to like A RELEASE

Committed variable(d)

- 15. Sketch a graph of the data for the experiment in Model 2. A space has been provided near to the data vable. Be more to opposite which variable belongs on each axis.
- 14. Consider the data in Model 2.
  - at. Which trials represent arbitions that are untattarered?

1.2. Identify the following variables in the experiment in Model 2:

No. 2 September

4. Which trial numbers represent substicces that are asturated?

and the other made bad, in-

a. Describe the feature in the graph that can help you identify the asturant solutions. Explain.

making the property of the second sec







# **Solubility Pogil Answer Key**

**L** Reisser

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Standard Test Method for Measurements of Aqueous Solubility, Standard Test Method for Solubility of Fixed Gases in Liquids, Solutions American Association for the Advancement of Science,1975 Construct a hypothesis to explain observations made or data collected concerning the effects of temperature and the nature of the solvent on the solubility of solids to demonstrate methods to manipulate and hold constant variables in a test of a hypothesis and to identify data that support and do not support a hypothesis Standard Test Method for Solubility of Fixed Gases in Low-Boiling Liquids, Solubility Data Series, 1996 Standard Test Methods for Estimation of Solubility of Gases in Petroleum and Other Organic Liquids ASTM International, 2012

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