

Solubility Rules Worksheet

1. Name or give the chemical formula for each of the following compounds.
2. State whether they are soluble (will dissolve) or insoluble (will not dissolve) in solution. Use solubility rules.

Chemical Formula	Name	Solubility
1. $\text{NH}_4\text{CH}_3\text{COO}$	Ammonium acetate	soluble
2. $\text{Ba}(\text{OH})_2$	Barium hydroxide	soluble
3. FeCO_3	Iron (II) Carbonate	insoluble
4. NaOH	Sodium hydroxide	soluble
5. RbNO_3	Rubidium nitrate	soluble
6. Cs_2SO_4	Cesium Sulfate	soluble
7. MgSO_4	Magnesium sulphate	soluble
8. ZnCl_2	Zinc chloride	soluble
9. $\text{Zn}(\text{OH})_2$	Zinc Hydroxide	insoluble
10. $\text{Zn}_3(\text{PO}_4)_2$	Zinc phosphate	insoluble
11. AgBr	Silver Bromide	insoluble
12. KNO_3	Potassium Nitrate	soluble
13. Al_2S_3	Aluminum Sulfide	insoluble
14. AgCH_3COO	Silver Acetate	insoluble
15. Sr_2CrO_4	Strontium Chromate	insoluble
16. AlPO_4	Aluminum Phosphate	insoluble
17. BaSO_4	Barium Sulfate	insoluble
18. $\text{Ca}(\text{OH})_2$	Calcium Hydroxide	insoluble
19. BaCO_3	Barium Carbonate	insoluble
20. MgCrO_4	Magnesium Chromate	soluble
21. Fe_2S_3	Iron (III) Sulfide	insoluble
22. NH_4CN	Ammonium Cyanide	soluble
23. AgI	Silver iodide	insoluble
24. Hg_2SO_4	Mercury (I) Sulfate	soluble
25. LiCl	Lithium Chloride	soluble

Solubility Rules Pogil Answers

U. Benedict



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