

Solutions Chemistry

Method of expressing	Formula
Mass percentage	$\left(\frac{w}{W}\right)\% = \frac{w_2}{(w_1 + w_2)} \times 100$
Volume percentage	$\left(\frac{v}{V}\right)\% = \frac{V_2}{(V_1 + V_2)} \times 100$
Mass by volume percentage	$\left(\frac{w}{V}\right)\% = \frac{w_2}{V_{\text{solution (in mL)}}} \times 100$
Mass fraction	$x_1 = \frac{w_1}{w_1 + w_2} \text{ or } x_2 = \frac{w_2}{w_1 + w_2}$
Strength (g L ⁻¹)	$\frac{w_2 \text{ (in g)}}{V_{\text{solution (in mL)}}} \times 1000$
Parts per million (ppm)	$\frac{w_2}{(w_1 + w_2)} \times 10^6$
Molarity, (M) (mol L ⁻¹)	$\frac{w_2 \times 1000}{M_2 \times V_{\text{solution (in mL)}}$
Molality, (m) (mol kg ⁻¹)	$\frac{w_2 \times 1000}{M_2 \times w_1 \text{ (in g)}}$
Mole fraction, (x)	$x_1 = \frac{n_1}{n_1 + n_2} \text{ or } x_2 = \frac{n_2}{n_1 + n_2},$ $(x_1 + x_2 = 1)$ In case of gases only, $y_1 = \frac{P_1}{P_1 + P_2} \text{ or } y_2 = \frac{P_2}{P_1 + P_2},$ $(y_1 + y_2 = 1)$

Solution For All Grade 12

Bruce D. Craig, David S. Anderson

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the authors' names.

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Anderson,1994-12-31 This book makes it easy for you to find what effect environment has on the corrosion of metals and alloys However this volume offers information on additional environments including concrete soil groundwater distilled water sodium acetate and more ThereAs also updated and expanded coverage of previously discussed environments as well as information on environments which deal with the dairy food brewing aerospace petrochemical and building industries The environments are listed alphabetically Each listing includes a general description of the conditions a comment on the corrosion characteristics of various alloys in such a situation a bibliography of recent articles specific to the environment tables consolidating and comparing corrosion rates at various temperatures and concentrations for various alloys and graphical information Also included are summaries on the general corrosion characteristics of major metals and alloys

Food Analysis Suzanne Nielsen,2003-04-30 This book provides information on the techniques needed to analyze foods in laboratory experiments All topics covered include information on the basic principles procedures advantages limitations and applications This book is ideal for undergraduate courses in food analysis and is also an invaluable reference to professionals in the food industry General information is provided on regulations standards labeling sampling and data handling as background for chapters on specific methods to determine the chemical composition and characteristics of foods Large expanded sections on spectroscopy and chromatography are also included Other methods and instrumentation such as thermal analysis selective electrodes enzymes and immunoassays are covered from the perspective of their use in the chemical analysis of foods A helpful Instructor s Manual is available to adopting professors **Turbomaths Grade 12** , **A**

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probability scenarios for students in grades 4 6 Students will encounter problems in which they read about students their age selling magazines for a school fund raiser concerned about their homeroom assignments and trying to decode the combination to a safe that their grandfather abandoned among others all of which maximizes learning so students gain a deep understanding of concepts in probability This book will help teachers parents and other educators to employ best

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