- (5 points) Table 4.8 (<u>Snoeyink</u> and Jenkins) shows the concentration of carbonate species for solutions with distilled water and 10-5M solutions of H₂CO₃*; sodium bicarbonate, NaHCO₃; and sodium carbonate, Na₂CO₃, in a system open to the atmosphere and in a closed system.
 - a. Why does the concentration of H₂CO₃* remain constant in the open system (across the different solutions) and decreases in a closed system?
 - b. For both systems, the concentration of H* decreases when comparing a solution of distilled water and 10-5M solutions of H2CO3*; NaHCO3 Na2CO3 with lowest concentration observed for Na2CO3 The decrease in H* concentration is greater for a closed system than for an open system. Explain this difference.

TABLE 4.8 Comparison of Composition and pH of Open and Closed Carbonate Systems

Solution	Closed System (Fig. 4-17)						
	[H*]	[OH-]	[H _i CO;]	[HCO,"]	[CO,*-]	pН	C1,00
10-4 M H,CO;	10-4.7	10-4.9	10-41	10-tr	10-16.4	5.7	10-54
10" M NaHCO,	10-7.4	10-4.4	10-44	10-1	10-54	7.6	10-5.0
10-4 M Na,CO,	10-*	10-4	10-7.4	10-5	10-4.0	9.0	10-44
Solution	Open System						
	[H*]	[OH-]	[H,CO;]	[HCO,-]	(CO,*-)	pН	C1.00,
10-4 M H,CO;	10-6.7	10-43	10-4	10-4.7	10-16.4	5.7	10-4.0
10" M NaHCO,	10-4.3	10 1.7	10-	10-1	10-44	6.3	10-4
10-4 M Na,00,	10-43	10-7.3	10-*	10-43	10 44	6.6	10-44

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Michael Rajnik

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Water & Sewage Works ,1973 Vols 76 include Reference and data section for 1929 1929 called Water works and sewerage data section Metals Speciation Separation and Recovery James W. Patterson,1987-07-05 This comprehensive book covers metals chemistry separation chemistry and metals separation processes State of the art papers give new and recent developments and future research needs Control of Disinfection By-products in Drinking Water Systems Anastasia Nikolau, Luigi Rizzo, Hüseyin Selcuk, 2007 The occurrence of disinfection by products DBPs in drinking water has been an issue of major concern during several decades The formation of many DBPs species during water disinfection has been documented while new by products are still being detected as the analytical instrumentation available becomes more accurate and sensitive Most of the DBPs have been proven to have toxic effects on living organisms therefore they pose risks to human health during drinking water consumption The factors affecting their formation have been extensively investigated their transport and fate have been studied modelling efforts for several of them have been performed in order to understand better their behaviour and therefore try to minimise their occurrence in waters Techniques for their

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