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**Zimbabwe School Examinations Council**

**GENERAL CERTIFICATE OF EDUCATION  
ORDINARY LEVEL**

# **INTEGRATED SCIENCE 5006**

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# Zimsec 2007 Intergrated Science Question Paper

**Hussin A.Rothana**



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web jun 30 2023 the limiting reagent is the one that is totally consumed it limits the reaction from continuing because there is none left to react with the in excess reactant there are two ways to determine the limiting reagent

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web a which chemical is the limiting reactant b how many grams of zns will be formed c how many grams of the excess reactant will remain after the reaction is over 3 which element is in excess when 3 00 grams of mg is ignited in 2 20 grams of pure oxygen what mass is in excess what mass of mgo is formed 4

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web reactant present in an amount lower than required by the reaction stoichiometry thus limiting the amount of product generated molecular equation chemical equation in which all reactants and products are represented as neutral substances

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