

Solutions Test Review

Key

Name:

Mrs. Z

Definitions:

F 1. solution

C 2. solute

G 3. solvent

A 4. soluble

B 5. aqueous solution

D 6. electrolyte

a. capable of being dissolved

b. solution with water as the solvent

c. substance that is dissolved in a solution

d. substance that dissolves in water to form a solution that conducts an electric current

f. homogeneous mixture of two or more substances in a single physical state

g. substance that does the dissolving in a solution

You must know:

7. List 3 ways you can increase the rate of dissolving.

↑ Stir, ↑ Temp, ↓ Particle Size

8. Explain what the phrase "like dissolves like" means in relation to solution formation.

Polar in Polar, Nonpolar in nonpolar

9. Explain how temperature affects the solubility of each of the following

a. solids

b. liquids

c. gases

↑ temp ↑ solubility

↑ temp ↓ solubility

10. Explain how pressure affects the solubility of each of the following

a. solids

b. liquids

c. gases

not affected

↑ pressure ↑ solubility

You must be able to solve problems similar to the following:

Molarity:

11. What mass of CH_3OH is required to prepare 1.50 liters of 3.00 M solution?

$$M = \frac{\text{mol}}{L} \quad \text{mol} = (M)(L) = (3.00)(1.50) = 4.5 \text{ mol } \text{CH}_3\text{OH}$$

$$\frac{4.5 \text{ mol } \text{CH}_3\text{OH}}{1 \text{ mol}} \times \frac{32.05 \text{ g}}{1 \text{ mol}} = \boxed{144 \text{ g } \text{CH}_3\text{OH}}$$

Solubility Review Answer Key

**Kai Matthes MD, PhD, Anjolie E.
Laubach MD, Ellen Wang MD, T.
Anthony Anderson MD**

Solubility Review Answer Key:

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