

Why?

Solubility of a salt depends upon the type of ions in the salt. Some salts are soluble in water and others are not. When two soluble salts are mixed together in water they may form a third insoluble salt. Net ionic equations are a way of showing the reactions that take place between two substances dissolved in water.

Learning Objectives

- Students will write simple net ionic equations for double displacement (replacement) reactions

Definitions

- An **aqueous** solution is a solution with water as the solvent.
- A **compound** is said to be soluble if it readily dissolves in water and does not precipitate if left undisturbed for an extended period of time.
- A **spectator ion** is an ion that is present during a reaction but does not take part in the reaction.
- A **net ionic equation** an equation that only shows the ions that undergo changes during a chemical reaction. (Spectator ions are omitted from net ionic equations.)

Model 1: Rules of Solubility in Aqueous Solutions

Several solid compounds are placed in water to determine if they are soluble. The results are shown in the chart below. An X indicates that the compound does not dissolve in water. If it does dissolve no mark is made. The top row shows the cation in the compound. The far left column shows the anion in the compound. For example: $\text{Mg}(\text{OH})_2$ is insoluble, MgBr_2 is soluble, AgCl is insoluble.

Table 1 - This table presents an overview of solubility of selected salts in water.

Table 1 - This table presents an overview of solubility of selected salts in water.

	NH_4^+	Li^+	Na^+	K^+	Mg^{2+}	Ca^{2+}	Sr^{2+}	Ba^{2+}	Ag^+	Pb^{2+}	Hg_2^{2+}	Fe^{2+}	Cu^{2+}	Zn^{2+}
NO_3^-														
$\text{C}_2\text{H}_3\text{O}_2^-$														
Cl^-									X	X	X			
Br^-									X	X	X			
I^-									X	X	X			
SO_4^{2-}						X	X	X	X	X				
OH^-					X				X	X	X	X	X	X
S^{2-}					X	X	X	X	X	X	X	X	X	X
CO_3^{2-}					X	X	X	X	X	X	X	X	X	X
PO_4^{3-}					X	X	X	X	X	X	X	X	X	X

- Is calcium carbonate soluble or insoluble?
- Is silver bromide soluble or insoluble?
- Is iron (III)sulfate soluble or insoluble?
- For what cations and anions are the compounds **always soluble** in water?
- For what anions are most of the compounds **usually soluble**?
 - For those anions that usually form soluble compounds, which cations result in the formation of insoluble compounds? List each cation separately.
- For what cations are most of their compounds **usually soluble**?

Label the following as soluble or nonsoluble.

- KOH
- CaSO_4
- NaNO_3
- FeS

Solubility Rules And Net Ionic Equations Pogil Activity Key

Lifelique

A decorative graphic consisting of a light blue horizontal bar with a rounded right end, and a red circular shape with a gradient, partially overlapping the bar's end.

Solubility Rules And Net Ionic Equations Pogil Activity Key:

Net Ionic Equations Lifeliqe,2019 This lesson plan covers the various ways to depict chemical equations using net ionic equations

Solubility Rules And Net Ionic Equations Pogil Activity Key: Bestsellers in 2023 The year 2023 has witnessed a noteworthy surge in literary brilliance, with numerous captivating novels enthralling the hearts of readers worldwide. Lets delve into the realm of top-selling books, exploring the captivating narratives that have captivated audiences this year. The Must-Read : Colleen Hoover's "It Ends with Us" This poignant tale of love, loss, and resilience has gripped readers with its raw and emotional exploration of domestic abuse. Hoover masterfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can prevail. Uncover the Best : Taylor Jenkins Reid's "The Seven Husbands of Evelyn Hugo" This captivating historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reid's captivating storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Solubility Rules And Net Ionic Equations Pogil Activity Key : Delia Owens' "Where the Crawdads Sing" This evocative coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens spins a tale of resilience, survival, and the transformative power of nature, captivating readers with its evocative prose and mesmerizing setting. These top-selling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of engaging stories waiting to be discovered. The novel begins with Richard Papan, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is an exceptional and suspenseful novel that will keep you speculating until the very end. The novel is a warning tale about the dangers of obsession and the power of evil.

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