Answers

Solids, Liquids, and Gases

Chapter Test A

- 1 B
- 2 C
- 3. A.
- 4. D
- 5. C
- 6. B
- 7. C
- 8. C
- 9. B
- 10 C
- 11 gas
- 12 thermal
- 13. gas
- 14. viscosity.
- 15. Boiling
- 16. F, more freely than
- 17 F. amorphous
- 18 F, freezing
- 19. F. sublimation
- 20 T
- 21. State B. The figure shows that the particles are arranged in a pattern of definite, fixed positions. They can vibrate in place but not move around.
- 22 State A. The particles are not in a fixed pattern as particles in a solid are, the particles can move around one another, but they remain quite close together.
- State C, because heating increases the energy of the particles. As the temperature of the liquid increases.

- the particles would begin to vibrate more quickly. They would move faster and begin to move farther apart.
- 2.4 In a gas, particles are spread far apart and they move in all directions at high speeds, filling all the space available. In a liquid, particles are packed closely together but can move freely around one another. As a result, a liquid can take on the shape of its container. In a solid, particles are packed tightly together, they can vibrate in place but not move from their fixed positions.
- 25. In an ice cube, water molecules are arranged in a crystalline form. Inside the freezer, the ice cube is at a temperature below 0°C. When it is removed from the freezer, the ice cube absorbs thermal energy from its surroundings, mostly from the air in the room. At first, the added thermal. energy causes the water molecules to vibrate faster, raising their temperature. When the temperature of the water molecules reaches 0°C, the melting point of water, the molecules vibrate so quickly that they break free from their fixed positions, and the ice begins to melt.
- 26. If the cylinder were heated, the temperature of the gas would increase. With the plunger held steady, the volume of the gas would remain constant. Under these conditions (increasing temperature and constant volume), the pressure of the gas would increase.

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William Thomas Stead

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