



PARTICLE THEORY AND STATES OF MATTER

What is the basic structure of matter? Understanding the nature of matter and the arrangement of particles is fundamental. The arrangement and movement of particles explain various properties of the matter.

Particle Theory is the idea that all matter is made up of tiny particles. These particles have three main characteristics:

- 1. Particles are very small.** They are too small to be seen with the naked eye. They can only be seen using a microscope.
- 2. Particles are constantly moving.** They are always in motion. The speed and direction of their motion is different in different states of matter.
- 3. Particles attract each other.** There are forces of attraction between particles, which vary depending on the type of matter, solids, liquids, and gases.

States of Matter Matter can exist in three primary states: solids, liquids, and gases. Each state has its own unique properties and arrangement of particles.

Solids: Particles are closely packed in a regular arrangement. They have a fixed shape and volume. The particles vibrate in fixed positions.

Liquids: Particles are closely packed but can move past each other. They have a fixed volume but no fixed shape. The particles move more freely than in solids.

Gases: Particles are far apart and move randomly. They have no fixed shape or volume. The particles move very fast in all directions.

Changes of State Matter can change from one state to another. These changes are called changes of state. They are caused by heating or cooling.

Heating: Adding heat to a substance causes its particles to move faster. This can lead to a change of state from solid to liquid (melting) or from liquid to gas (boiling).

Cooling: Removing heat from a substance causes its particles to move slower. This can lead to a change of state from gas to liquid (condensation) or from liquid to solid (freezing).



11. How does the arrangement of particles change when a solid melts? What happens to the shape of the solid? Can you give examples of substances that do this?

10. Compare the arrangement of particles in each state of matter. How does this affect the properties of the change of state?

9. What is the change of state, and what does it mean?

8. Why is the property of matter a substance that its physical properties are different? Give examples of the different states of matter.

7. How does the property of matter change when it changes?

1. What is particle theory?

2. List all the states of matter. Give an example of each state.

3. How does the arrangement of particles change when a solid melts? What does this suggest to you?

4. The solid state is the most stable state of matter. Why is this? Give an example of a substance that is in the solid state.

5. What are the three states of matter?

6. How does the property of matter change when it changes?

States Of Matter Guided Reading Answer Key

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States Of Matter Guided Reading Answer Key:

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