# Nuclear Chemistry – Alpha, Beta, and Gamma Decay Practice

Identify the following examples as alpha, beta, or gamma decay.

1. 
$$^{233}_{92}U \rightarrow ^{4}_{2}He + ^{229}_{90}Th _alpha_$$

4. 
$$^{202}_{81}\text{TI} \rightarrow ^{202}_{81}\text{TI} + ^{0}_{0}\text{Y}$$
 gamma

2. 
$$^{99}_{43}$$
Tc  $\rightarrow ^{0}_{-1}$ e +  $^{99}_{44}$ Ru beta

s. 
$${}^{125}_{53}I \rightarrow {}^{0}_{0} \Upsilon + {}^{125}_{53}I$$
 gamma

3. 
$$^{175}_{78}$$
 Pt  $\rightarrow ^{171}_{76}$  Os +  $^{4}_{2}$  He \_alpha\_

6. 
$${}^{82}_{35}Br \rightarrow {}^{82}_{36}Kr + {}^{0}_{-1}e$$
 \_\_beta\_\_\_

Alpha Decay: Write the alpha decay equations for the following nuclides. Example:  ${}^{208}_{84}$  Po  $\rightarrow {}^{4}_{2}$  He  ${}^{204}_{82}$  Pb

Example: 
$${}^{208}_{84}$$
 Po  $\rightarrow {}^{4}_{2}$  He +  ${}^{204}_{82}$  Pb

7. 
$$^{256}_{103}$$
Lr  $\rightarrow ^{4}_{2}$ He +  $^{252}_{101}$ Md

s. 
$$^{231}_{91}$$
 Pa  $\rightarrow ^{4}_{2}$  He +  $^{227}_{89}$  Ac

Beta Decay: Write the beta decay equations for the following nuclides.

Example: 
$${}^{14}_{6}C \rightarrow {}^{14}_{7}N + {}^{0}_{-1}e$$

11. 
$$^{24}_{11}$$
 Na  $\rightarrow ^{24}_{8}$  O +  $^{0}_{-1}$  e

13. 
$$^{247}_{95}$$
 Am  $\rightarrow ^{247}_{96}$  Cm +  $^{0}_{-1}$  e

12. 
$$^{201}_{79}$$
 Au  $\rightarrow ^{201}_{80}$  Hg +  $^{0}_{-1}$  e

14. 
$$^{90}_{38}$$
Sr  $\rightarrow ^{90}_{39}$ Y +  $^{0}_{19}$ e

Balancing Decay Reactions: Fill in the blank in each of the following decay reactions with the correct decay particle or decayed nucleus that will balance the reaction, also state whether it is alpha or beta decay

15. 
$$^{184}_{74}$$
 W  $\rightarrow ^{4}_{2}$  He +  $^{180}_{72}$  Hf

20. 
$$^{60}_{27}W \rightarrow ^{0}_{-1}e + ^{60}_{28}Ni$$

**K Morrison** 

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#### **Table of Contents Standardized Test Practice Nuclear Chemistry Solution Manual**

- 1. Understanding the eBook Standardized Test Practice Nuclear Chemistry Solution Manual
  - The Rise of Digital Reading Standardized Test Practice Nuclear Chemistry Solution Manual
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Standardized Test Practice Nuclear Chemistry Solution Manual
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Standardized Test Practice Nuclear Chemistry Solution Manual
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Standardized Test Practice Nuclear Chemistry Solution Manual
  - Personalized Recommendations
  - Standardized Test Practice Nuclear Chemistry Solution Manual User Reviews and Ratings
  - Standardized Test Practice Nuclear Chemistry Solution Manual and Bestseller Lists
- 5. Accessing Standardized Test Practice Nuclear Chemistry Solution Manual Free and Paid eBooks
  - Standardized Test Practice Nuclear Chemistry Solution Manual Public Domain eBooks
  - Standardized Test Practice Nuclear Chemistry Solution Manual eBook Subscription Services
  - Standardized Test Practice Nuclear Chemistry Solution Manual Budget-Friendly Options
- 6. Navigating Standardized Test Practice Nuclear Chemistry Solution Manual eBook Formats

- o ePub, PDF, MOBI, and More
- Standardized Test Practice Nuclear Chemistry Solution Manual Compatibility with Devices
- Standardized Test Practice Nuclear Chemistry Solution Manual Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - o Adjustable Fonts and Text Sizes of Standardized Test Practice Nuclear Chemistry Solution Manual
  - Highlighting and Note-Taking Standardized Test Practice Nuclear Chemistry Solution Manual
  - Interactive Elements Standardized Test Practice Nuclear Chemistry Solution Manual
- 8. Staying Engaged with Standardized Test Practice Nuclear Chemistry Solution Manual
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Standardized Test Practice Nuclear Chemistry Solution Manual
- 9. Balancing eBooks and Physical Books Standardized Test Practice Nuclear Chemistry Solution Manual
  - Benefits of a Digital Library
  - o Creating a Diverse Reading Collection Standardized Test Practice Nuclear Chemistry Solution Manual
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Standardized Test Practice Nuclear Chemistry Solution Manual
  - $\circ \ \ Setting \ Reading \ Goals \ Standardized \ Test \ Practice \ Nuclear \ Chemistry \ Solution \ Manual$
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Standardized Test Practice Nuclear Chemistry Solution Manual
  - Fact-Checking eBook Content of Standardized Test Practice Nuclear Chemistry Solution Manual
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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