

SS 2 CHEMISTRY PRACTICAL 3RD TERM EXMINATION

1. B is a solution containing 5.30g of anhydrous sodium trioxocarbonate (IV) per dm³. A is a solution of hydrochloric acid of unknown concentration.
Put A into the burette and titrate 25cm³ (or 20cm³) portions of B with A. Using methyl orange as the indicator. Record the volume of your pipette.
Tabulate your burette readings, and calculate the average volume of acid used.
From your results, calculate:
 - a. The concentration of solution B in moles per dm³
 - b. The concentration of solution A in moles per dm³
 - c. The mass concentration in g per dm³ of the acid in solution A

The equation of reaction is :



(H = 1; C = 12; O = 16; Na = 23; Cl = 35.5)

2. (a). If you were provided with anhydrous CaCO₃, spatula and stirrer (i) list three other materials you would require to prepare a standard solution of CaCO₃ (ii) State what you would observe on adding dilute HCl to a portion of the CaCO₃
(b). State what would be observed when aqueous ammonia is added to: (i) Litmus paper; (ii) Pb(NO₃)₂ solution in drops until in excess.

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Xiang Xie



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Eco-efficient Repair and Rehabilitation of Concrete Infrastructures Fernando Pacheco-Torgal, Robert E. Melchers, Xianming Shi, Nele de Belie, Kim Van Tittelboom, Andres Saez Perez, 2017-11-15 Eco efficient Repair and Rehabilitation of Concrete Infrastructures provides an updated state of the art review on eco efficient repair and rehabilitation of concrete infrastructure The first section focuses on deterioration assessment methods and includes chapters on stress wave assessment ground penetrating radar monitoring of corrosion SHM using acoustic emission and optical fiber sensors Other sections discuss the development and application of several new innovative repair and rehabilitation materials including geopolymer concrete sulfoaluminate cement based concrete engineered cementitious composites ECC based concrete bacteria based concrete concrete with encapsulated polyurethane and concrete with super absorbent polymer SAPs amongst other topics Final sections focus on crucial design aspects such as quality control including lifecycle and cost analysis with several related case studies on repair and rehabilitation The book will be an essential reference resource for materials scientists civil and structural engineers architects structural designers and contractors working in the construction industry Delivers the latest research findings with contributions from leading international experts Provides fully updated information on the European standard on materials for concrete repair EN 1504 Includes an entire sections on the state of the art in NDT innovative repair and rehabilitation materials as well as LCC and LCA information **Sustaining Protein Nutrition Through Plant-Based Foods: A Paradigm Shift** Sapna Langyan, Pranjal Yadava, Tarun Belwal, Tanushri Kaul, 2022-11-25 **Directory [of] Officers, Faculty, and Staff and Associated Organizations** University of Michigan, 1995 *Advanced Practical Chemistry* Jagdamba Singh, 2009 *Practical Chemistry (For B.Sc. I, II and III Year Students)* Pandey O.P. & Bajpai D.N. & Giri S., 1972 FOR B Sc I II III YEAR STUDENTS Chemistry Practical Cape Technikon School of Physical Sciences, 198? **Practical Chemistry** Eric John Holmyard, 1923 College Practical Chemistry, 2010 **College Practical Chemistry**, 2010 **Comprehensive Practical Chemistry XI** Dr. N. K. Verma, B. K. Vermani, Dr. Neera Verma, 2010-02 **Course of Practical Chemistry** Matthew Moncrieff Pattison Muir, 1899 **Chemistry (Practical A).**, 1946 **Course of Practical Chemistry ...** William Odling (M.B.), 1865 Key to Chemistry Graham Chadwick Hill, 1978 **Chemistry (Practical Examinations).**, 1932 **Comprehensive Practical Chemistry (Abhilekhan) XII** Dr. N.K. Verma, Avinash Ojha, 2008-05-01 A Course of Practical Chemistry, 1746 **Keys to Chemistry.**, 1978 **Practical/Laboratory Manual Chemistry Class - XI** Er. Meera Goyal, 2021-05-29 1 Basic Laboratory Techniques 1 To cut a glass tube or glass rod 2 To bend the glass rod at an angle 3 To draw a glass jet from a glass tube 4 To bore a cork and fit a glass tube into it Viva Voce 2 Characterisation and Purification of Chemical Substances 1 To determine the melting point of the given unknown organic compound and its identification simple laboratory technique Viva Voce 2 To determine the boiling point of a given liquid when available in small quantity simple laboratory method Viva

Voce 3 To prepare crystals of pure potash alum $K_2SO_4 \cdot Al_2(SO_4)_3 \cdot 24H_2O$ from the given impure sample 4 To prepare the pure crystals of copper sulphate from the given crude sample 5 To prepare pure crystals of benzoic acid from a given impure sample Viva Voce 3 Measurement of pH Values 1 To determine the pH value of vegetable juices fruit juices tap water and washing soda by using universal pH paper 2 To determine and compare the pH values of solutions of strong acid HCl and weak acid CH_3COOH of same concentration 3 To study the pH change in the titration of strong base Vs strong acid by using universal indicator paper 4 To study the pH change by common ion CH_3COO^- ion in case of weak acid CH_3COOH 5 To determine the change in pH value of weak base NH_4OH in presence of a common ion NH_4^+ Viva Voce 4 Chemical Equilibrium 1 To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions 2 To study the shift in equilibrium between $Co(H_2O)_6^{2+}$ and Cl^- ions by changing the concentrations of either of the ions Viva Voce 5 Quantitative Analysis 1 To prepare M 10 oxalic acid solution by direct weighing method 2 To prepare M 10 solution of sodium carbonate by direct weighing method 3 To determine the strength of given solution of sodium hydroxide by titrating it against N 10 or M 20 solution of oxalic acid 4 To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N 10 or M 20 sodium carbonate solution Viva Voce 6 Qualitative Analysis Analysis of Anions Analysis of Cations Viva Voce 7 Detection of Elements in Organic Compounds 1 To detect the presence of nitrogen sulphur and halogens in a given organic compound by Lassaigne's test 2 To detect the presence of nitrogen sulphur and halogens in the given organic compound sample number by Lassaigne's test Viva Voce INVESTIGATORY PROJECTS 1 Checking of Bacterial Contamination in Water 1 To check the bacterial contamination in drinking water by testing sulphide ions Viva Voce 2 Methods of Water Purification 1 To purify water from suspended impurities by using sedimentation 2 To purify water by boiling 3 To purify water by distillation method 4 To purify water by reverse osmosis technique 5 To purify water by GAC method 6 To purify water by bleach treatment 7 To purify water by oxidising agent 8 To purify water by ozone treatment method Viva Voce 3 Water Analysis 1 To test the hardness of different water samples Viva Voce 4 Foaming Capacity of Various Soaps 1 To compare the foaming capacity of different washing soaps 2 To study the effect of addition of sodium carbonate on foaming capacity of washing soap Viva Voce 5 Tea Analysis 1 To study the acidity of different samples of tea leaves tea by using pH paper Viva Voce 6 Analysis of Fruits and Vegetable Juices 1 To analysis the fruit and vegetable juices for the constituent present in them Viva Voce 7 Rate of Evaporation 1 To study the rate of evaporation of different liquids IViva Voce 8 Effect of Acids and Bases on Tensile Strength of Fibres 1 To compare the tensile strength of natural fibres and synthetic fibres 2 To study the effect of acids and bases on tensile strength of different fibres Viva Voce

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